

Equilibrium Constant - Worksheet 1_ANSWERS

1. Write the equilibrium expression for the following reactions:



$$K = \frac{[U]^u [V]^v}{[W]^w [X]^x}$$



$$K = \frac{[H_2][CO_2]}{[H_2O][CO]}$$



$$K = \frac{[Cl_2][CO]}{[COCl_2]}$$



$$K = \frac{[HCl]^2}{[H_2][Cl_2]}$$



$$K = \frac{[NO][CO_2]}{[NO_2][CO]}$$



$$K = \frac{[Zn^{2+}]}{[Ag^+]^2}$$

*Note: solids are not included



$$K = \frac{[H_2][C_2H_4]}{[C_2H_6]}$$

2. Consider the following equilibrium reaction:



The K_{eq} for this reaction at 25°C is 1.02. At equilibrium the concentration of HBr is 0.50 mol/L. Assuming H_2 and Br_2 are present in equal amounts, calculate the concentration of H_2 at equilibrium.

$$[H_2] = 0.50 \text{ mol/L}$$

Equilibrium Unit Webquest And Notes Answer Key

DJ Losen



Equilibrium Unit Webquest And Notes Answer Key:

This book delves into Equilibrium Unit Webquest And Notes Answer Key. Equilibrium Unit Webquest And Notes Answer Key is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Equilibrium Unit Webquest And Notes Answer Key, encompassing both the fundamentals and more intricate discussions.

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- Chapter 1: Introduction to Equilibrium Unit Webquest And Notes Answer Key
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- Chapter 3: Equilibrium Unit Webquest And Notes Answer Key in Everyday Life
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- Chapter 5: Conclusion

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