

# TITRATIONS



Calculate the concentration of the sulfuric acid in mol/dm<sup>3</sup>.

Use only the student's concordant results.

mean titre =  $\frac{22.10 + 22.15 + 22.15}{3} = 22.13 \text{ cm}^3$  ✓

$$n(\text{NaOH}) = 0.105 \times (22.13/1000) = 0.002324 \text{ mol} \quad \checkmark$$

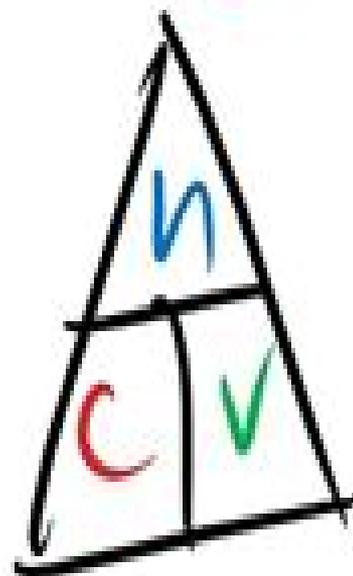
$$n(\text{H}_2\text{SO}_4) = 0.002324 / 2 = 0.001162 \text{ mol} \quad \checkmark$$

$$c(\text{H}_2\text{SO}_4) = \frac{0.001162}{(25/1000)} = 0.0465 \text{ mol/dm}^3 \quad \checkmark$$

$$n = m/M_r$$

$$n = C \times V$$

$$C = n/V$$



# GCSE CHEMISTRY

# Gcse Titration Practise Questions

**Guillaume Favre**



## Gcse Titration Practise Questions:

## Gcse Titration Practise Questions Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Gcse Titration Practise Questions**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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and share the learning outcomes of civics in elementary school fifth grade students the research was conducted in kebon  
baru sdn 10 pagi by the sample is fifth grade students in the second semester of academic year 2016

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