



PRE-LAB QUESTIONS

1. Compare and contrast diffusion and osmosis.
 - *Osmosis is the process of solvent traveling from a dilute solution to a concentrated solution to equalize the concentration*
 - *Diffusion is the process of particles moving from an area of high concentration to an area of lower concentration to equalize the concentration*
 - *Both processes have an end goal of solution concentration equalization*
 - *Both processes involve moving from an area of high concentration to an area of lower concentration (in osmosis, the high concentration of water (the dilute) is the high concentration)*
 - *Osmosis commonly refers to water molecules, where diffusion can be any mixture*

2. Draw a picture of a cell in isotonic, hypotonic, and hypertonic states.



3. Why don't red blood cells swell or shrink in blood?
 - *Red blood cells do not shrink or swell in blood because blood and the red blood cells are in an isotonic state*
4. How do osmotic power plants work?

Osmotic power plants capture energy through osmosis of fresh and salt water. When the salt water and fresh water are mixed energy is released. A process of pressure retarded osmosis (PRO) captures the released energy during separation.
5. Research the structures that protect plant and animal cells from damage resulting from osmotic pressure. Write a few paragraphs explaining what they are, how they work, and where they are located.



Escience Lab Osmosis Answers

Christian Drosten



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