

ATOMS

READING COMPREHENSION

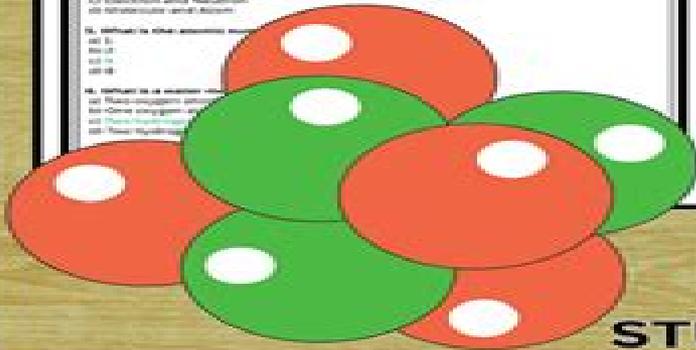
Name: _____

Questions:

1. What are atoms?
 - a) Large blocks of matter
 - b) Tiny building blocks of matter
 - c) A type of chemical reaction
 - d) Tools used in science labs
2. Which of the following particles has a negative charge?
 - a) Proton
 - b) Neutron
 - c) Electron
 - d) Neutrino
3. What is found at the center of an atom?
 - a) Electrons
 - b) Neutrons
 - c) Protons and Neutrons
 - d) Neutrinos
4. Which two particles are found in the nucleus?
 - a) Electrons and Proton
 - b) Proton and Neutron
 - c) Electron and Neutron
 - d) Neutrino and Atom
5. What is the atomic number of carbon?
 - a) 2
 - b) 6
 - c) 12
 - d) 14
6. What is a water molecule made up?
 - a) Two oxygen atoms and one hydrogen atom
 - b) One oxygen atom and one hydrogen atom
 - c) Two hydrogen atoms and one oxygen atom
 - d) Two hydrogen atoms and two oxygen atoms

Answer Key:

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ATOMS

There are so many things in this world. They come in different colors, sizes, and shapes. But did you know that all those things are made up of the same material? Everything we see around us is made up of **atoms**.



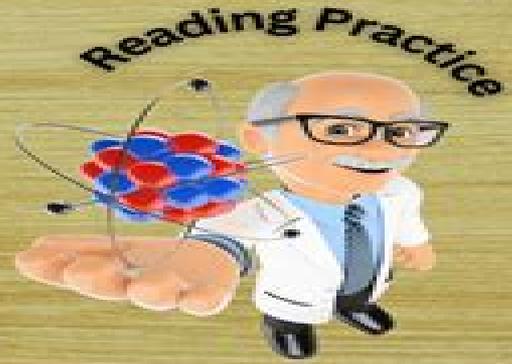
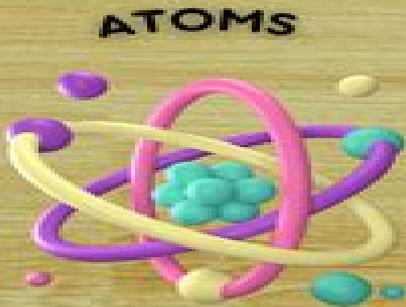
Have you seen a house made up of bricks? Atoms are the smallest building blocks of matter, in the same way bricks are the building blocks of a house. Stars, planets, animals are all made of atoms. Your body and your notebook are both made of atoms!

An atom is very small. We can't see an atom with our naked eyes. Scientists use special instruments to observe an atom.

Inside an atom there are three types of particles: Electron, Proton and Neutron. Protons and Neutrons together make the **Nucleus** at the centre of an atom. Electrons move around the nucleus, sort of like the planets orbit the Sun.

The electrons carry a negative charge and protons carry a positive charge. Neutrons are neutral, means they don't carry any charge. Opposite charges attract each other, so protons and electrons attract each other, just like two magnets.

There are many different types of atoms. Have you heard of elements like Iron, Gold and Carbon? Each different kind of atom is a different element. Different atoms have different number of protons inside them. Number of protons in an atom is called its Atomic number. Hydrogen atom has one proton, so its atomic number is one. A carbon atom has 6 protons, its atomic number is 6. There are 118 different elements known to us. Scientists have organised all the elements in a table called Periodic Table. Two or more atoms combine together to form a molecule. A water molecule is made of two atoms of Hydrogen and one atom of Oxygen.



Introduction To Atoms Elementary Reading And Study Answers

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