

Acids and Bases

Section 19.1 Acids and Bases: An Introduction

In your textbook, read about the properties of acids and bases.

For each description below, write *acid* if it tells about a property of an acid or *base* if it tells about a property of a base. If the property does not apply to either an acid or a base, write *neither*. If it applies to both an acid and a base, write *both*.

- _____ 1. Can turn litmus paper a different color
- _____ 2. Reacts with certain metals
- _____ 3. Contains more hydrogen ions than hydroxide ions
- _____ 4. Feels slippery
- _____ 5. Reacts with carbonates
- _____ 6. Feels rough
- _____ 7. Contains equal numbers of hydrogen and hydroxide ions
- _____ 8. Tastes bitter
- _____ 9. Tastes sour

In your textbook, read about the different models of acids and bases.

Use the terms below to complete the passage. You may use each term more than once.

Arrhenius conjugate base	Bronsted-Lowry hydrogen	conjugate acid hydroxide
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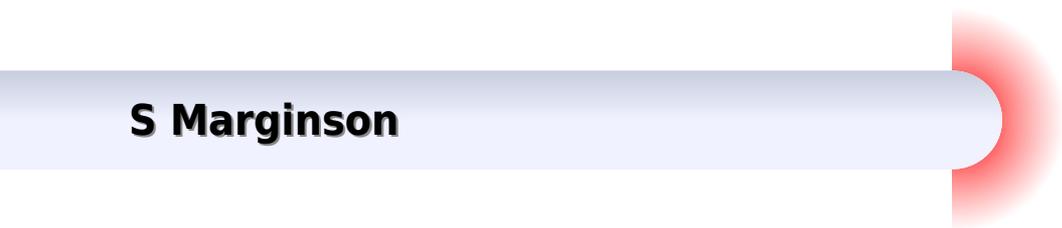
The **(10)** _____ model of acids and bases states that an acid contains the element **(11)** _____ and forms ions of this element when it is dissolved in water. A base contains the **(12)** _____ group and dissociates to produce **(13)** _____ ions in aqueous solution.

According to the **(14)** _____ model, an acid donates **(15)** _____ ions, and a base accepts **(16)** _____ ions. According to this model, in an acid-base reaction, each acid has a **(17)** _____, and each base has a **(18)** _____.

Introduction To Acids And Bases Worksheets Chapter 18

1

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