

ELEMENTS COMPOUNDS AND MIXTURES



Chemical structure: CCN(C)C(=O)C

Elements and Compounds
Mixtures vs. Compounds

Directions: Define the following terms and give an example of each in a separate sheet of paper. Write 10 for mixtures and 10 for compounds.

1. Mixture: Mixture
 Answer: Mixture

2. Pure substance: Pure substance
 Answer: Pure substance

3. Element: Element
 Answer: Element

4. Compound: Compound
 Answer: Compound

5. Carbon Dioxide (CO₂): Carbon Dioxide
 Answer: Carbon Dioxide

6. Water (H₂O): Water
 Answer: Water

7. Salt (NaCl): Sodium Chloride
 Answer: Sodium Chloride

8. Sugar (C₆H₁₂O₆): Sugar
 Answer: Sugar

Name: _____ Class: _____

Elements and Compounds

Directions:
 1. Identify whether each given item is an element or a compound.
 2. If it is an element, write down its chemical symbol.
 3. If it is a compound, write down its chemical formula.

Ex. Oxygen

Type	Chemical Symbol
Element	O

1. Water

Type	Chemical Symbol

2. Carbon Dioxide

Type	Chemical Symbol

Chemical structure: CCN(C)C

Elements and Compounds

Element

Directions:

- 1. Define: A pure substance made up of only one type of atom.
- 2. Examples: Hydrogen (H₂), Oxygen (O₂), Carbon (C), Nitrogen (N₂), Gold (Au).

Directions:

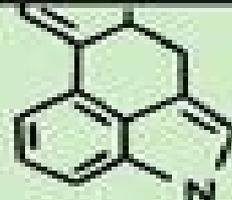
- 1. Definition: A substance composed of two or more different elements combined in a fixed ratio.
- 2. Examples: Water (H₂O), Carbon Dioxide (CO₂), Sugar (C₆H₁₂O₆).

Examples of Elements:

- Hydrogen (H₂)
- Oxygen (O₂)
- Carbon (C)
- Nitrogen (N₂)
- Gold (Au)



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Sabine Zange



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