

ANSWER KEY

ATOMS AND ELECTRONS STUDY GUIDE

Learning Goals: (1) (10-1) P. 41 - Explain a new scientific theory of atomic-scale processes (atomic theory) by describing changes in the atomic model over time, including those changes that are consistent with experimental evidence. (10-1) P. 42 - Explain the relative physical and chemical behavior of atoms by describing the structure of atoms in terms of positive nuclear charge, neutral particles, and negative charges that are in motion. (10-1) P. 42 - Explain properties of atoms and their changes in the periodic table in terms of periodicity of their properties. (10-1) P. 42 - Explain the importance of atoms in the fundamental phenomena of chemistry, physics, astronomy, and other fields using matter.

ANSWERS

Question: Match the atomic theory element to its correct conclusion.

Conclusion	Atomic Model	Atomic Theory Element
1. Electrons are made of atoms.	Democritus (460 BC)	A. Everything is made of atoms. (Democritus) (no correct choice)
2. Matter is made of particles.	Leucippus (450 BC)	B. Matter is made of particles. (Leucippus) (no correct choice)
3. Atoms are indivisible particles.	Thomson (1897)	C. Atoms are indivisible. (Thomson)
4. Atoms have a positive charge.	Rutherford (1911)	D. Atoms have a positive charge. (Rutherford)
5. Electrons are present in atoms.	Rutherford (1911)	E. Electrons are present in atoms. (Rutherford)
6. Atoms are mostly empty space.	Rutherford (1911)	F. Atoms are mostly empty space. (Rutherford)

Question: Use the periodic table to answer the following questions:

1. **Atomic number** = **P** = **15** - The number of protons equals the number of electrons.
2. **Atomic number** minus the **atomic number** will equal the **number of neutrons**.
3. **P** = **15** - **10** = **5** - **5** protons are balanced by **5** electrons (the rest are charged either **positively** or **negatively**).

Electrons In Atoms Study Guide Answer Key

John A. Olmsted, Gregory M. Williams

Electrons In Atoms Study Guide Answer Key:

Student Study Guide to accompany Alcamo's Fundamentals of Microbiology Jeffrey C. Pommerville, 2010-08-10

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