

Materials Science and Engineering Final Exam Review 2023

What is a crystalline solid? - ANSWER Crystalline solids are atoms, ions, or molecules that make up a solid in a regular, well-defined arrangement.

What is a Unit Cell? - ANSWER The smallest repeating pattern of crystalline solids

(Unit cells are like bricks in a wall—they are all identical and repeating)

There are 4 types of Crystalline solids, what are they? - ANSWER Ionic Solids

Molecular Solids

Covalent Solids

Metallic Solids.

Ionic Solids - ANSWER Positive and negative ions and held together by electrostatic attractions.

Very high melting points

brittleness

poor conductors

(An example of an ionic solid is table salt, NaCl)

Molecular Solids - ANSWER Atoms or molecules held together by London dispersion forces, dipole-dipole forces, or hydrogen bonds.

-low melting points

-flexibility

-poor conductors

(Example: Ice)

Introduction To Engineering Final Exam Review

DP Hallahan



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