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# Lsengine Service Manual

**G Psacharopoulos**



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web donor whereas a base generically represented by b is a proton acceptor when acids and bases combine they swap a proton  $HA + B \rightleftharpoons HB + A^-$  strong acids e.g. sulfuric acid have a very pronounced tendency to transfer a proton to a base whereas weak acids

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web understand that parts of an acid base titration be able to determine the  $K_a$  or  $K_b$  from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate  $K_{a1}$  and  $K_{a2}$  for a polyprotic acid by the end of this lab students should be able to

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web solution a titrant volume 0 ml the solution ph is due to the acid ionization of  $\text{HCl}$  because this is a strong acid the ionization is complete and the hydronium ion molarity is 0.100 M the ph of the solution is then  $\text{pH} = \log 0.100 = 1.000$   $\text{pH} = \log 0.100 = 1.000$  b titrant volume 12.50 ml

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