



Purpose

The purpose of this experiment is to determine the oxidation numbers of elements in free atoms, ions, and compounds, write formulas based on ionic charge, name and write formulas for a wide array of binary compounds, and name and write formulas for a wide array of ternary compounds.

Learning Objectives

- Define the term oxidation number or oxidation state.
- Define and correctly apply the terms anion and cation to various charged particles.
- Assign oxidation numbers for elements in uncombined states, charged states, or compounds by applying the rules for oxidation numbers.
- Write formulas for compounds based on ionic charges.
- Name and write formulas for binary compounds of metals and nonmetals.
- Name and write formulas for binary compounds of nonmetals and metalloids.
- Name and write formulas for binary acids.
- Name and write formulas for ionic compounds containing polyatomic ions.
- Name and write formulas for ternary acids.
- Name and write formulas for acid salts.

Laboratory Skills

None

Equipment

- None

Chemicals

- None



Formulas And Nomenclature Oxidation Lab

Clemens Wendtner



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