

General Chemistry

Chapter 2

★ Quantum Mechanics

- Theory using observations as an explanation for something we can NOT see
- Absolutely small matter behaved MUCH differently than a matter of the macroscopic world.
- Subatomic particles appeared to exist in two conditions they had a natural duality
 - EX: How you act individually is different than how you act in a crowd
- Much of the behavior of atoms is determined by the electron
 - Because the electrons are the only subatomic particles NOT in the nucleus

★ Nature of Light

- Light is a form of electromagnetic radiation
- All electromagnetic waves move through space at the same constant speed
 - FORMULA on slides
- Amplitude is a measure of light intensity- the larger the amplitude, the brighter the light
- Wavelength is a measure of the distance covered by the wave
- It's 1 point on a wave to the same point on the next wave
- λ + Frequency and Wavelengths are inverses of each other as one gets bigger the other gets smaller

★ Color

- The color of light is determined by its wavelength or frequency
- White light is a mixture of all the colors of visible light
 - It's a spectrum
 - Red, orange, yellow, green, blue, indigo, violet
- When an object absorbs some of the wavelengths of white light and reflects others, it appears colored; the observed color is predominantly the colors reflected.
- Frequency
 - The number of waves that pass a certain point in a given period of time
 - The number of waves = the number of cycles
- Total Energy-
 - The proportional amplitude of the waves and their frequency
 - The larger the amplitude the more force it has
 - The more frequent the wave strikes, the more force there is
- Long wave - low frequency
- Short wave - high frequency

★ Electromagnetic Spectrum

- Visible light is 400 to 700 nm of the electromagnetic spectrum
- Radio wave light has the lowest energy

General Chemistry Chapter 2nuclear

KJ Lindholm-Leary



General Chemistry Chapter 2 nuclear:

This book delves into General Chemistry Chapter 2nuclear. General Chemistry Chapter 2nuclear is a crucial topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into General Chemistry Chapter 2nuclear, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to General Chemistry Chapter 2nuclear
 - Chapter 2: Essential Elements of General Chemistry Chapter 2nuclear
 - Chapter 3: General Chemistry Chapter 2nuclear in Everyday Life
 - Chapter 4: General Chemistry Chapter 2nuclear in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, the author will provide an overview of General Chemistry Chapter 2nuclear. This chapter will explore what General Chemistry Chapter 2nuclear is, why General Chemistry Chapter 2nuclear is vital, and how to effectively learn about General Chemistry Chapter 2nuclear.
 3. In chapter 2, this book will delve into the foundational concepts of General Chemistry Chapter 2nuclear. The second chapter will elucidate the essential principles that must be understood to grasp General Chemistry Chapter 2nuclear in its entirety.
 4. In chapter 3, this book will examine the practical applications of General Chemistry Chapter 2nuclear in daily life. The third chapter will showcase real-world examples of how General Chemistry Chapter 2nuclear can be effectively utilized in everyday scenarios.
 5. In chapter 4, this book will scrutinize the relevance of General Chemistry Chapter 2nuclear in specific contexts. The fourth chapter will explore how General Chemistry Chapter 2nuclear is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, the author will draw a conclusion about General Chemistry Chapter 2nuclear. This chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of General Chemistry Chapter 2nuclear.

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