



CHEMISTRY

CHEAT SHEET



NAMING SIMPLE COMPOUNDS

NAMING SIMPLE COMPOUNDS CHEAT SHEET

Naming Compounds with Two Elements
 (Name the more electronegative element first)

- EXAMPLE**
- CO2
 Carbon Dioxide (Naming Rule #1)
- Directions for naming:**
1. Name the more electronegative element first.
 - a. Name the element as is.
 - b. Name the element with a prefix (see below) to show the number of atoms.
 2. Name the less electronegative element as the second element.
 - a. Name the element with a prefix (see below) to show the number of atoms.
 - b. Name the element as is.
 3. Name the compound with the name of the less electronegative element first and the name of the more electronegative element second.

Prefixes	
Number	Prefix
1	mono-
2	di-
3	tri-
4	tetra-
5	penta-
6	hexa-
7	hepta-
8	octa-
9	nona-
10	deca-

Suffixes	
Number	Suffix
1	-ide
2	-diide
3	-tride
4	-tetride
5	-pentide
6	-hexide
7	-heptide
8	-octide
9	-nonide
10	-decide

Naming Compounds with a Metal and a Nonmetal

- EXAMPLE**
- CaCl2
 Calcium Chloride
- Directions for naming:**
1. Name the metal as is.
 2. Name the nonmetal with a prefix (see below) to show the number of atoms. Name the nonmetal as if it were a metal.

Prefixes			
Number	Prefix	Number	Prefix
1	mono-	1	mono-
2	di-	2	di-
3	tri-	3	tri-
4	tetra-	4	tetra-
5	penta-	5	penta-
6	hexa-	6	hexa-
7	hepta-	7	hepta-
8	octa-	8	octa-
9	nona-	9	nona-
10	deca-	10	deca-

Common Names for Specific Compounds			
Formula	Common Name	Formula	Common Name
<chem>NaCl</chem>	Salt	<chem>CaCl2</chem>	Calcium Chloride
<chem>NaOH</chem>	Soda	<chem>CaCO3</chem>	Calcium Carbonate
<chem>Na2CO3</chem>	Soda Ash	<chem>CaSO4</chem>	Calcium Sulfate
<chem>NaHCO3</chem>	Baking Soda	<chem>Ca(OH)2</chem>	Slaked Lime
<chem>NaNO3</chem>	Saltpeter	<chem>Ca(OH)2</chem>	Slaked Lime

Naming Compounds Study Guide

S Marginson



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Table of Contents Naming Compounds Study Guide

1. Understanding the eBook Naming Compounds Study Guide
 - The Rise of Digital Reading Naming Compounds Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Naming Compounds Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Naming Compounds Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Naming Compounds Study Guide
 - Personalized Recommendations
 - Naming Compounds Study Guide User Reviews and Ratings
 - Naming Compounds Study Guide and Bestseller Lists
5. Accessing Naming Compounds Study Guide Free and Paid eBooks

- Naming Compounds Study Guide Public Domain eBooks
 - Naming Compounds Study Guide eBook Subscription Services
 - Naming Compounds Study Guide Budget-Friendly Options
6. Navigating Naming Compounds Study Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Naming Compounds Study Guide Compatibility with Devices
 - Naming Compounds Study Guide Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Naming Compounds Study Guide
 - Highlighting and Note-Taking Naming Compounds Study Guide
 - Interactive Elements Naming Compounds Study Guide
 8. Staying Engaged with Naming Compounds Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Naming Compounds Study Guide
 9. Balancing eBooks and Physical Books Naming Compounds Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Naming Compounds Study Guide
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Naming Compounds Study Guide
 - Setting Reading Goals Naming Compounds Study Guide
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Naming Compounds Study Guide
 - Fact-Checking eBook Content of Naming Compounds Study Guide
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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