

## Limiting Reactant Worksheet #2

1. Given the following reaction: (Balance the equation first!)



- If you start with 14.8 g of  $\text{C}_3\text{H}_8$  and 3.44 g of  $\text{O}_2$ , determine the limiting reactant.
- Determine the number of moles of carbon dioxide produced.
- Determine the number of grams of  $\text{H}_2\text{O}$  produced.
- Determine the number of grams of excess reactant left.

2. Given the following equation:



- If 10.0 g of  $\text{Al}_2(\text{SO}_4)_3$  is reacted with 10.0 g of  $\text{NaOH}$ , determine the limiting reactant.
- Determine the number of moles of  $\text{Al}(\text{OH})_3$  produced.
- Determine the number of grams of  $\text{Na}_2\text{SO}_4$  produced.
- Determine the number of grams of excess reactant left over in the reaction.

3. Given the following equation:



- If 25.4 g of  $\text{Al}_2\text{O}_3$  is reacted with 10.2 g of  $\text{Fe}$ , determine the limiting reactant.
- Determine the number of moles of  $\text{Al}$  produced.
- Determine the number of grams of  $\text{Fe}_3\text{O}_4$  produced.
- Determine the number of grams of excess reactant left over in the reaction.

4. Consider the reaction



- 80.0 grams of  $\text{I}_2\text{O}_5$ , reacts with 28.0 grams of carbon monoxide,  $\text{CO}$ . Determine the mass of iodine  $\text{I}_2$ , which could be produced?
- If, in the above situation, only 0.160 moles, of iodine,  $\text{I}_2$  was produced.
  - What mass of iodine was produced?
  - What percentage yield of iodine was produced?

5. Zinc and sulfur react to form zinc sulfide according to the equation



If 25.0 g of zinc and 30.0 g of sulfur are mixed,

- Which chemical is the limiting reactant?
- How many grams of  $\text{ZnS}$  will be formed?
- How many grams of the excess reactant will remain after the reaction is over?

6. Which element is in excess when 3.00 grams of  $\text{Mg}$  is ignited in 2.20 grams of pure oxygen? What mass is in excess? What mass of  $\text{MgO}$  is formed?

7. How many grams of  $\text{Al}_2\text{S}_3$  are formed when 5.00 grams of  $\text{Al}$  is heated with 10.0 grams  $\text{S}$ ?

# Limiting Reagent Worksheet 202 Answer Key

**Yan Bai**



**Limiting Reagent Worksheet 202 Answer Key:**

## Decoding **Limiting Reagent Worksheet 202 Answer Key**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Limiting Reagent Worksheet 202 Answer Key**," a mesmerizing literary creation penned with a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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