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2 TRANSFERENCIA DE CALOR

que indicaría una transferencia neta del calor, es por un valor bajo de la capacidad térmica ρc . Un valor bajo en la capacidad térmica podría significar que se absorbe mayor cantidad de energía de la que se emite por el material y se usa para elevar la temperatura del material, así se dispondría de más energía para transferir. La difusividad térmica y los números de Biot son considerados por separado.

En las ecuaciones anteriores, la expresión de la derivada en x , de ser la escribo en la forma de desarrollo de Taylor habiendo tomado solo los dos primeros términos de este desarrollo.

La Ecu. (1.3a) puede transformarse a coordenadas cilíndricas o esféricas mediante técnicas normales del cálculo. Los resultados son los siguientes:

Coordenadas cilíndricas:

$$\frac{\partial^2 T}{\partial x^2} + \frac{1}{r} \frac{\partial T}{\partial r} + \frac{1}{r^2} \frac{\partial^2 T}{\partial \phi^2} + \frac{\partial^2 T}{\partial z^2} + \dot{q} = \frac{1}{\alpha} \frac{\partial T}{\partial t} \quad (1.3b)$$

Coordenadas esféricas:

$$\frac{1}{r^2} \frac{\partial}{\partial r} (r^2 \frac{\partial T}{\partial r}) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial}{\partial \phi} \left(\sin^2 \theta \frac{\partial T}{\partial \phi} \right) + \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial T}{\partial \theta} \right) + \frac{\partial^2 T}{\partial z^2} + \dot{q} = \frac{1}{\alpha} \frac{\partial T}{\partial t} \quad (1.3c)$$

Los sistemas de coordenadas para el uso de las Ecs. (1.3b) y (1.3c) se indican en las Figuras 1.3b y 1.3c, respectivamente.

Muchos problemas prácticos implican sólo casos especiales de las ecuaciones generales dadas anteriormente. Como guía de los desarrollos en los capítulos siguientes, se dará mostrar la forma reducida de las ecuaciones generales para algunos casos de interés práctico.

Flujo de calor estacionario y unidimensional sin generación de calor:

$$\frac{\partial^2 T}{\partial x^2} = 0 \quad (1.4)$$

Nótese que esta ecuación es la misma que la Ecu. (1.1) cuando $\dot{q} =$ constante.

Flujo de calor estacionario y unidimensional sin generación de calor en coordenadas cilíndricas:

$$\frac{\partial^2 T}{\partial x^2} + \frac{1}{r} \frac{\partial T}{\partial r} = 0 \quad (1.5)$$

Flujo de calor estacionario y unidimensional con fuente de calor:

$$\frac{\partial^2 T}{\partial x^2} + \dot{q} = 0 \quad (1.6)$$

Conducción estacionaria y bidimensional de fuente de calor:

$$\frac{\partial^2 T}{\partial x^2} + \frac{\partial^2 T}{\partial y^2} = 0 \quad (1.7)$$

1.2. CONDUCTIVIDAD TÉRMICA

La Ecu. (1.1) es la que define la conductividad térmica. Basándose en esta definición pueden medirse mediante experimentos para determinar la conductividad térmica de diferentes materiales. Para gases, a temperaturas moderadamente bajas, pueden utilizarse los tratamientos analíticos de la teoría cinética de gases para predecir con precisión los valores observados experimentalmente. En algunos casos, se dispone de teorías para la predicción de las conductividades térmicas de líquidos y sólidos, pero, por lo general, cuando se trata de líquidos y sólidos es preciso establecer algunas ecuaciones y conceptos basados en hechos.

El mecanismo de la conducción térmica en gases es muy simple. Se imagina la energía cinética de una molécula con un momento; así, en una región de alta temperatura, las moléculas poseen velocidades más altas que en una región de baja temperatura. Las moléculas están en constante movimiento aleatorio, chocando unas con otras e intercambiando energía y cantidad de movimiento. Las moléculas tienen un movimiento aleatorio, existe o no un gradiente de temperatura en el gas. Si una molécula se mueve desde una región de alta temperatura a otra de menor temperatura, transporta energía cinética hacia la zona del sistema de baja temperatura y cada vez que energía mediante los choques con las moléculas de menor energía.

En la Tabla 1.1 se da la lista de valores típicos de la conductividad térmica de algunos materiales para indicar los órdenes de magnitud relativos que se esperan en la práctica. En el Apéndice A se da una tabla con información más completa. En general, la conductividad térmica depende fuertemente de la temperatura.

Se imagina que la conductividad térmica tiene unidades de calor por metro y por grado Celsius cuando el flujo de calor se expresa en valores. Nótese que está involucrada la capacidad del calor y el calor mismo de la conductividad térmica indica los ritmos que el calor fluye en un material dado. ¿Cómo se ha tratado en cuanto la capacidad de la transferencia de energía en el medio molecular del que se ha hablado anteriormente?

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Adrian Bejan



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