

Matter and Change, pp. 1-4

- | | | |
|----|----|---|
| 2. | 5. | c |
| 3. | 5. | c |
| 4. | 7. | c |
| 9. | 3. | a |
- mass and volume
 - group or family
 - homogeneous
 - products
 - metal
 - solid state
 - chemical properties
 - atom
 - compound
 - nonmetal
 - mixture
 - pure substance
 - basic
 - metalloid
 - reactants: carbon and oxygen; product: carbon dioxide
 - reactant: mercury(II) oxide; products: mercury and oxygen
 - f
 - g
 - c
 - a
 - b
 - i
 - d
 - h
 - e
 - physical change
 - chemical change
 - physical change
 - physical change
 - chemical change
 - physical change
 - In a solid, particles are packed together in relatively fixed positions. The particles vibrate about a fixed point. In a liquid, particles are close together but can flow around one another. In a gas, particles are at great distances from one another, compared to the particles of liquids and solids.
 - The composition of a pure substance is the same throughout and does not vary from sample to sample. Pure substances include elements and compounds. A mixture contains more than one substance and can vary in composition from one sample to another and sometimes within different parts of the same sample.
 - Although energy can be absorbed or released in a

change, it is not destroyed or created. It simply takes a different form.

- Homogeneous mixtures are uniform in composition; heterogeneous mixtures are not.
- An element cannot be broken down, but water can be decomposed into hydrogen and oxygen by passing an electrical current through it (electrolysis).

Measurements and Calculations, pp. 5-8

- | | | | |
|----|----|----|---|
| a | 5. | c | |
| 2. | c | 6. | c |
| 3. | c | 7. | a |
| 4. | a | 8. | a |
- time
 - mass
 - density
 - energy
 - length or distance
 - volume
 - area
 - qualitative
 - quantitative
 - qualitative
 - quantitative
 - 3.00×10^7 km/s
 - three
 - 0.026 g or 2.6×10^{-2} g
 - 2.5×10^{-4} L
 - quantity
 - derived units
 - conversion factor
 - 0.0432 kg or 4.32×10^{-2} kg
 - 5400 mL or 5.4×10^3 mL
 300. K
 - 1.05×10^3 J
 - 3.51×10^3 cm
 - A hypothesis is a testable statement that can be used to make predictions and to carry out further experiments. A theory is a broad generalization that explains a body of facts or phenomena.
 - Mass is a measure of the quantity of matter. Weight is a measure of the gravitational pull on matter. Mass does not depend on gravitational attraction.
 - Models are visual, verbal, or mathematical representations. They are used to explain how phenomena occur or how data or events are related.
 - The graph of two quantities that are directly proportional is a straight line through the origin.

Glencoe Science Chemistry Matter Change Answer Key Chapter 2

Ensheng Dong



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Unveiling the Magic of Words: A Review of "**Glencoe Science Chemistry Matter Change Answer Key Chapter 2**"

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