

Name: _____

Period: _____



- Solute **A**
 - Solvent **B**
 - Covalent **C**
 - Ionic **D**
 - Concentration **E**
- M** or **N**

- The part that dissolves into the solution.
- The part that dissolves something.
- Compounds made up of two non-metals.
- Since they are made up of ions, they are polar molecules.
- Tells you how much solvent is dissolved in the solution.

- Dissolve **E**
- Pressure **A**
- Crushing **C**
- Heating **D**
- Stirring **B**

- Causes more gaseous solvent to be forced into a liquid.
- Increases solubility by moving the molecules around faster.
- Increases solubility by allowing the solvent to touch more sides of the solute.
- Expands the solvent a bit to allow more room for the solute.
- When something seems to disappear inside a solvent.

- Water tension **F**
- Dissociation **A**
- Hydrogen bonds **D**
- Capillary action **B**
- Cohesion **C**

- The breaking up of an ionic molecule by a polar solvent.
- A molecule that has a positive and negative end.
- The attractive force between two molecules of the same substance.
- The attractive force caused by hydrogen protons to a more negative non-metal.
- What allows plants to draw water up from their roots.
- Caused by the cohesion of water molecules holding it together.

- Water is polar or non-polar?
- Cooking oil **N**
 - Soluble in water **P**
 - A compound with H's **P/N**
 - Sugar **P/N**

- Salt **P**
- Ionic Compounds **P**
- Dissolves in water **N**
- Insoluble in water **N**

Draw a water molecule, being sure to label the ends as positive and negative.



Give three ways to make sugar dissolve faster.

- stirring
- heating
- crushing

Do metals become positive or negative?

Would a metal be attracted to water's hydrogen or oxygen?

Would a nonmetal be attracted to water's hydrogen or oxygen?

Which is the smallest particle size: rock salt, regular salt, or only powdered salt?

Which is the largest particle size: sugar cube, granulated sugar (regular table sugar), or powdered sugar?

To which side of a Water Molecule are these Attracted?

\ominus Magnesium	\ominus Sulfur	\ominus Iodine
\oplus Chlorine	\oplus Potassium	\oplus Bromine
\oplus Calcium	neither	Helium
	(candy)	Nitrogen

Which one will dissolve the fastest?

powdered sugar or granulated sugar?

hot water or cold water?

stirred or not stirred?

Large particles or small particles?

 OR 

Which is warmer: arctic water or tropical water?

Which holds more oxygen: arctic water or tropical water?

When power plants use cold streams for cooling, they can kill the fish by heating up the water. Why do many fish die? lack of oxygen (warm water causes the O_2 to be released in air)

Which holds more solid solute: hot or cold liquids?

Which holds more gaseous solute: hot or cold liquids?

Why are water bugs able to "walk on water"?

water tension (cohesion)

How do plants get water from their roots up to their leaves?

capillary action

What property of gases allows them to be pressured into solution?

can be compressed

Gym Packet Answer Key

Albert A Gayle



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Table of Contents Gym Packet Answer Key

1. Understanding the eBook Gym Packet Answer Key
 - The Rise of Digital Reading Gym Packet Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Gym Packet Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gym Packet Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gym Packet Answer Key
 - Personalized Recommendations
 - Gym Packet Answer Key User Reviews and Ratings
 - Gym Packet Answer Key and Bestseller Lists
5. Accessing Gym Packet Answer Key Free and Paid eBooks
 - Gym Packet Answer Key Public Domain eBooks

- Gym Packet Answer Key eBook Subscription Services
- Gym Packet Answer Key Budget-Friendly Options
- 6. Navigating Gym Packet Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Gym Packet Answer Key Compatibility with Devices
 - Gym Packet Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gym Packet Answer Key
 - Highlighting and Note-Taking Gym Packet Answer Key
 - Interactive Elements Gym Packet Answer Key
- 8. Staying Engaged with Gym Packet Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gym Packet Answer Key
- 9. Balancing eBooks and Physical Books Gym Packet Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gym Packet Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gym Packet Answer Key
 - Setting Reading Goals Gym Packet Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gym Packet Answer Key
 - Fact-Checking eBook Content of Gym Packet Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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