

Gas Laws Graphing Activity

- In this activity you will graph three different graphs, and then answer questions about your graphs.
- Use at least half a sheet of graph paper to make your graph. You may use the back. Graph paper is available at the front desk.
- Make sure to label the x-axis, y-axis, and place a title on your graph.
- **You will connect the points on the graph using colored pencils. They are located on the side table.**

I. Pressure vs. Volume

Volume (L)	Pressure (atm)
1.20	0.500
0.600	1.00
0.300	2.00
0.200	3.00
0.150	4.00
0.120	5.00
0.100	6.00

Volume should be placed on the x-axis, and pressure placed on the y-axis.

II. Temperature vs. Volume

Temperature (K)	Volume (L)
700.	1.400
601	1.202
499	0.998
401	0.802
300.	0.600
199	0.398
100.	0.200

Temperature should be placed on the x-axis, and volume placed on the y-axis.

III. Pressure vs. Temperature

Temperature (K)	Pressure (atm)
100.	0.2632
150.	0.3947
200.	0.5263
300.	0.7895
400.	1.0526
450.	1.1842

Gas Laws Graphing Activity Answers

RJ Shavelson



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