

Q1

The common oxidation states of Ti are

- (a) +2 and +3 (b) +3 and +4
(c) -3 and -4 (d) +2, +3 and +4

Q2

When phenol is treated with excess bromine water, it gives

- (a) *m*-Bromophenol (b) *o*- and *p*-Bromophenols
(c) 2,4-Dibromophenol (d) 2,4,6-Tribromophenol.

Q3

If 0.5 mol of BaCl_2 is mixed with 0.2 mole of Na_3PO_4 , find the maximum amount of $\text{Ba}_3(\text{PO}_4)_2$ that can be formed.

- (a) 1 mole (b) 0.5 mole (c) 0.1 mole (d) 0.01 mole

Q4

Which is chemically most active non-metal?

- (a) S (b) O (c) F (d) N

Q5

The following species will not exhibit disproportionation reaction

- (a) ClO^- (b) ClO_2^- (c) ClO_3^- (d) ClO_4^-

Q6

Which of the following has been used in the manufacture of non-inflammable photographic films?

- (a) Cellulose nitrate (b) Cellulose xanthate
(c) Cellulose perchlorate (d) Cellulose acetate

Q7

Point out the wrong statement:

For a first order reaction

- (a) time for half-change ($t_{1/2}$) is independent of initial concentration
(b) change in the concentration unit does not change the rate constant (k)
(c) time for half-change \times rate constant = 0.693
(d) the unit of k is $\text{mol}^{-1} \text{min}^{-1}$

Q8

Name of the structure of silicates in which three oxygen atoms of $[\text{SiO}_4]^{4-}$ are shared.

- (a) Pyrosilicate
(b) Sheet silicate
(c) Linear chain silicate
(d) Three dimensional silicate

Q9

In the following sequence of reactions



- (a) *n*-propyl amine (b) isopropylamine
(c) ethylamine (d) ethylmethyl amine

Q10

Which of the following monosaccharide is a pentose?

- (a) Galactose (b) Glucose
(c) Fructose (d) Arabinose

Q11

In Bohr's model, atomic radius of the first orbit is y , the radius of the 3rd orbit, is

- (a) $y/3$ (b) y
(c) $3y$ (d) $9y$

Q12

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