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# G K93725 Jun13 Ch3hp

**Jicheng Xie**



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q 1 what are the characteristics of solid state ans i they have definite mass volume and shape ii intermolecular distances are short iii intermolecular forces are strong iv their constituent particles atoms molecules or ions have fixed positions and can

only oscillate about their mean positions

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the solid state class 12 cbse revision notes this article deals with solid state notes first of all people call liquids and gases as fluids this is due to their ability to flow furthermore this fluidity is possible because their molecules are free to move about

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according to the cbse syllabus 2023 24 this chapter has been removed from ncert class 12 chemistry textbook 1 solids solids have definite volume shape and mass due to the short distance between the fixed position of particles and the strong interactions between them 1 1 characteristic properties of the solid state

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chapter 1 the solid state talks about two main categories of solids which are crystalline solids and amorphous solids before understanding the differences between these solids let us learn their characteristic properties they have a definite shape volume and mass

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there are three states of matter solid liquid and gas in chapter 1 of ncert solutions for class 12 chemistry solid characteristics are studied solids are defined by their mass shape and volume intermolecular distance and force are both short yet intermolecular force is strong