

1. According to the Bronsted—Lowry definition, which chemical species can function *both* as an acid and as a base?
- Cl^-
 - SO_4^{2-}
 - NH_4^+
 - HCO_3^-
 - H_3O^+
2. In the reaction $\text{CN}^- + \text{H}_2\text{O} \rightleftharpoons \text{HCN} + \text{OH}^-$ which is an acid—base conjugate pair?
- H_2O and HCN
 - H_2O and OH^-
 - CN^- and H_2O
 - HCN and OH^-
3. Given that HX is a stronger Bronsted acid than HY in aqueous solution, which is true of a 1 M solution of NaX ?
- It is less basic than a 1 M solution of NaY .
 - It is more basic than a 1 M solution of NaY .
 - It yields a neutral solution.
 - It is more concentrated than a 1 M solution of NaY .
4. The pH of a 0.03 M HCl solution is
- 1.5
 - 2.5
 - 3.5
 - 12.5
5. The pH of a 1.0×10^{-2} M $\text{Ba}(\text{OH})_2$ solution at 25°C is
- 2.7
 - 3.0
 - 11.0
 - 11.3
6. Which is the strongest acid?
- HClO
 - HClO_2
 - HClO_3
 - HClO_4
7. Which salt reacts with water (hydrolyzes) to produce a basic solution?
- $\text{NaC}_2\text{H}_3\text{O}_2$
 - NaNO_3
 - NH_4Cl
 - BaSO_4

8. The weakest of the bases listed is

Acid	Conjugate Base	K_a (Ionization Constant of Acid)
HCl	Cl^-	100% ionized
HSO_4^-	SO_4^{2-}	1.2×10^{-2}
H_2S	HS^-	5.7×10^{-8}
HS^-	S^{2-}	1.2×10^{-14}

- Cl^-
 - CN^-
 - HS^-
 - S^{2-}
 - SO_4^{2-}
9. The oxide of which element will react with water to form the strongest acid?
- | | Main Groups | | | | | | | |
|--------|-------------|----|-----|----|---|----|-----|---|
| Period | I | II | III | IV | V | VI | VII | O |
| First | | | | | | | | |
| Second | X | Y | Z | P | | Q | S | U |
| Third | W | | | | | R | T | M |
- W
 - M
 - P
 - R
 - Z
10. Which statement is a logical inference from the fact that a 0.10 M solution of potassium acetate $\text{KC}_2\text{H}_3\text{O}_2$ is less alkaline than a 0.10 M solution of potassium cyanide, KCN ?
- Hydrocyanic acid is a weaker acid than acetic acid.
 - Hydrocyanic acid is less soluble in water than acetic acid.
 - Cyanides are less soluble than acetates.
 - Acetic acid is a weaker acid than hydrocyanic acid.
11. In the titration of 50.0 mL of 0.100 M benzoic acid (a monoprotic acid) with 50.0 mL of 0.100 M NaOH , the properties of the solution at the equivalence point will correspond exactly to the properties of
- a 0.100 M sodium solution.
 - a 0.0500 M sodium hydroxide solution.
 - a 0.0500 M benzoic acid solution.
 - a 0.0500 M sodium benzoate solution.

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web details typical lead time 4 6 weeks the wide open design evolution rock crawler chassis was initially designed for trail riding and competing in various forms of off road racing strength space and style were all thoroughly thought out when designing this chassis the demand came for a smaller hill killer style chassis like this one

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web step 2 building it rope bridge measure wheel base of crawler mine is 310mm x 430mm working with this size i decided to make bridge the length of a pallet and about 70mm wider narrowing to 20mm smaller then my crawler mark this across five boards and cut with saw i then ripped each board into 15mm strips then lay strips out on

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web may 27 2021 you can go off roading to some degree with almost any 4x4 vehicle under the sun but if you re expressly wanting to go rock crawling then there are some things you really need to

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web jun 17 2015 as well a five point harness has more mounting points so your weight is more evenly distributed across the harness rock crawler with aluminum doors a set of aluminum doors will keep you and

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web so how do you build an rc rock crawler course gather the materials and supplies lay down the plastic strips lay down the rocks create another rock layer add another layer of rocks landscaping optional test out the rc rock crawler course