

Naming Ionic Compounds Practice Worksheet - Solutions

Name the following ionic compounds:

- | | | |
|-----|------------------------------|--------------------------------|
| 1) | NH_4Cl | ammonium chloride |
| 2) | $\text{Fe}(\text{NO}_3)_3$ | iron (III) nitrate |
| 3) | TiBr_3 | titanium (III) bromide |
| 4) | Cu_3P | copper (I) phosphide |
| 5) | SnSe_2 | tin (IV) selenide |
| 6) | GaAs | gallium arsenide |
| 7) | $\text{Pb}(\text{SO}_4)_2$ | lead (IV) sulfate |
| 8) | $\text{Be}(\text{HCO}_3)_2$ | beryllium bicarbonate |
| 9) | $\text{Mn}_2(\text{SO}_3)_3$ | manganese (III) sulfite |
| 10) | $\text{Al}(\text{CN})_3$ | aluminum cyanide |

Write the formulas for the following compounds:

- | | | |
|-----|--------------------------|---|
| 11) | chromium (VI) phosphate | $\text{Cr}(\text{PO}_4)_2$ |
| 12) | vanadium (IV) carbonate | $\text{V}(\text{CO}_3)_2$ |
| 13) | tin (II) nitrite | $\text{Sn}(\text{NO}_2)_2$ |
| 14) | cobalt (III) oxide | Co_2O_3 |
| 15) | titanium (II) acetate | $\text{Ti}(\text{C}_2\text{H}_3\text{O}_2)_2$ |
| 16) | vanadium (V) sulfide | V_2S_5 |
| 17) | chromium (III) hydroxide | $\text{Cr}(\text{OH})_3$ |
| 18) | lithium iodide | LiI |
| 19) | lead (II) nitride | Pb_3N_2 |
| 20) | silver bromide | AgBr |

Naming Ionic Compounds Practice Answer Key

JL Elias



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