

# IONIC BONDING 1

Name: \_\_\_\_\_

Ionic bonding occurs between a metal and a non-metal. In the chart below show the transfer of electrons with an arrow.



1. $\text{K} + \text{O}$	$\text{K} \cdot + \cdot \ddot{\text{O}} \cdot \rightarrow \text{K}^+ \text{O}^{2-}$	$\text{K}_2\text{O}$
2. $\text{Mg} + \text{O}$	$\text{Mg} \cdot + \cdot \ddot{\text{O}} \cdot \rightarrow \text{Mg}^{2+} \text{O}^{2-}$	$\text{MgO}$
3. $\text{Ca} + \text{S}$	$\text{Ca} \cdot + \cdot \ddot{\text{S}} \cdot \rightarrow \text{Ca}^{2+} \text{S}^{2-}$	$\text{CaS}$
4. $\text{Na} + \text{Br}$	$\text{Na} \cdot + \cdot \ddot{\text{Br}} \cdot \rightarrow \text{Na}^+ \text{Br}^-$	$\text{NaBr}$
5. $\text{H} + \text{Cl}$	$\text{H} \cdot + \cdot \ddot{\text{Cl}} \cdot \rightarrow \text{H}^+ \text{Cl}^-$	$\text{HCl}$
6. $\text{Al} + \text{Cl}$	$\text{Al} \cdot + 3 \cdot \ddot{\text{Cl}} \cdot \rightarrow \text{Al}^{3+} \text{Cl}_3^{3-}$	$\text{AlCl}_3$
7. $\text{N} + \text{O}$	$3 \cdot \ddot{\text{N}} \cdot + 2 \cdot \ddot{\text{O}} \cdot \rightarrow 2 \text{N}^{3-} \text{O}^{2+}_3$	$\text{N}_2\text{O}_3$
8. $\text{Cr} + \text{S} + \text{S}$	$2 \text{Cr} \cdot + 3 \cdot \ddot{\text{S}} \cdot \rightarrow 2 \text{Cr}^{3+} \text{S}_3^{2-}$	$\text{Cr}_2\text{S}_3$

# Ionic Bonding Instructional Fair Answer Key

**Marcel A. Müller**



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