

Name: \_\_\_\_\_

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Chemistry 11

**Classifying Matter**

Assignment

1. Complete the following sentences by filling in the appropriate word from the list below.

Elements

Mixtures

Compound

Symbols

Properties

Carbon

Pure

Letters

Carbon dioxide

- a. Elements cannot be separated by physical or chemical means.
- b. Elements are abbreviated with symbols, which consist of one or two letters.
- c. Two or more elements chemically combined make up a compound.
- d. Carbon is an example of an element and carbon dioxide is an example of a compound.
- e. Elements and compounds are called pure substances because they have a unique set of chemical and physical properties.

2. Fill in the blanks in the following chart.

Element Name	Element Symbol	Common Ions Formed
a. Sodium	<u>Na</u>	<u>+1</u>
b. Copper	<u>Cu</u>	<u>+2, +1</u>
c. Tungsten	<u>W</u>	<u>+6</u>
d. Iron	<u>Fe</u>	<u>+3, +2</u>
e. Tin	<u>Sn</u>	<u>+4, +2</u>
f. Potassium	<u>K</u>	<u>+1</u>
g. Gold	<u>Au</u>	<u>+3, +1</u>
h. Silver	<u>Ag</u>	<u>+1</u>

3. Classify each of the following as one of an Atom (A), a Molecule (M) or an Ion (I).

- a.  $P^{3-}$  I
- b.  $H_2O$  M
- c.  $O_2$  M
- d.  $Au^{3+}$  I

- e. Ge A
- f. O A
- g.  $Ca^{2+}$  I
- h.  $NH_3$  M

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**SA Dillow**



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