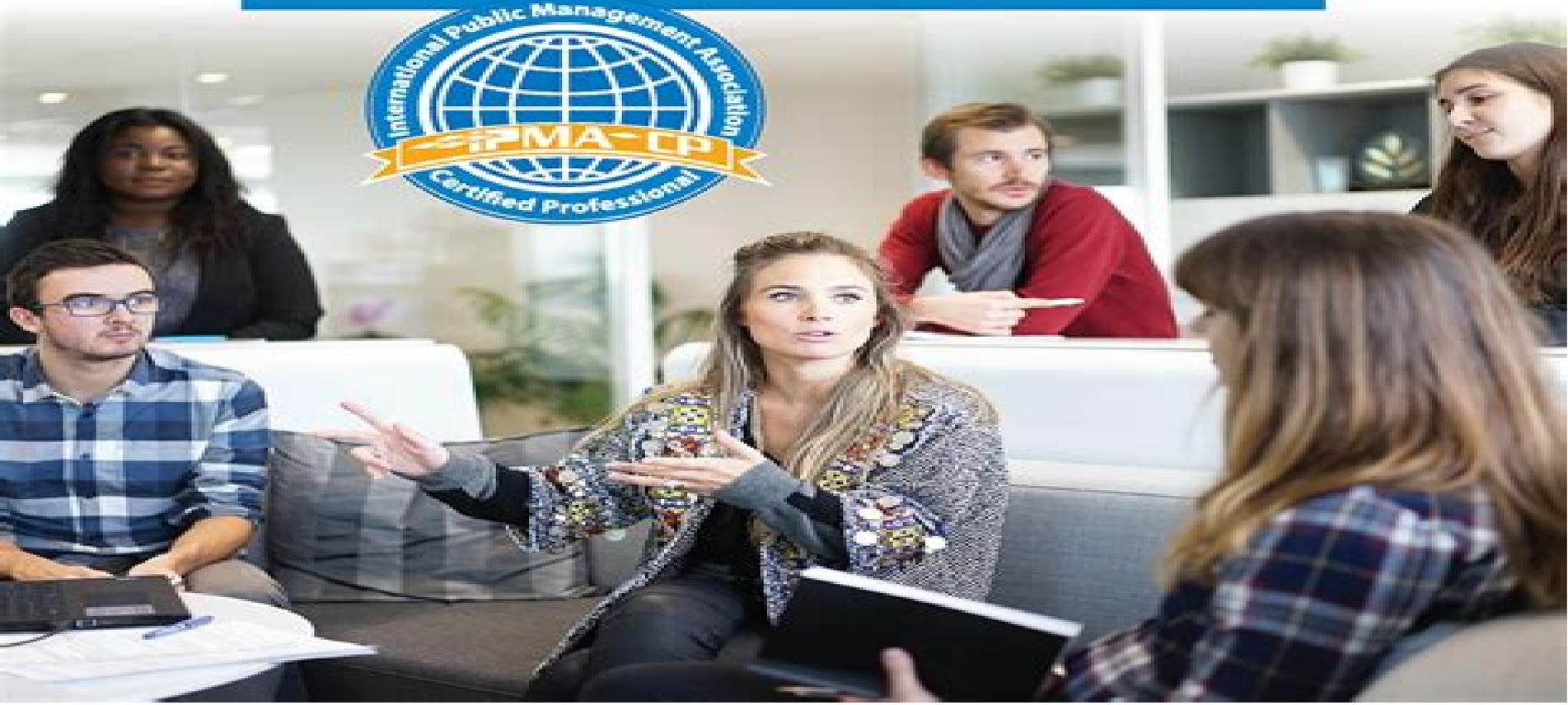




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web solution a titrant volume 0 ml the solution pH is due to the acid ionization of  $\text{HCl}$  because this is a strong acid the ionization is complete and the hydronium ion molarity is 0.100 M the pH of the solution is then  $\text{pH} = \log 0.100 = 1.000$  pH log 0.100 = 1.000 b titrant volume 12.50 ml

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