



Introduction to working as a Chemist.

Use a highlighter as you work through this booklet. ESL students, make vocab lists of the terminology that is defined or described here.

Assessment Statements

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| <ul style="list-style-type: none"> <input type="checkbox"/> Outline the Scientific Method <input type="checkbox"/> Identify independent, dependent & controlled variables. <input type="checkbox"/> Demonstrate correct and safe use of laboratory equipment. <input type="checkbox"/> Distinguish between quantitative and qualitative data <ul style="list-style-type: none"> > <i>Qualitative is descriptive, not using numbers, e.g. properties, colour, gas production, sound, odor.</i> > <i>Quantitative is mathematical, including</i> <ul style="list-style-type: none"> ↳ <i>Raw data (as recorded)</i> ↳ <i>Processed (after calculations)</i> <input type="checkbox"/> Distinguish between accuracy and precision <ul style="list-style-type: none"> > <i>Accuracy describes the correctness of a set of measurements</i> > <i>Precision describes the repeatability of a set of measurements</i> | <ul style="list-style-type: none"> <input type="checkbox"/> State SI units of measurement for <ul style="list-style-type: none"> > <i>Mass: grams (g)</i> > <i>Length: Metres (m), millimeters (mm)</i> > <i>Concentration: molarity (M)</i> > <i>Temperature: Degrees Celsius (°C)</i> > <i>Volume: Litres (L) or millilitres (mL)</i> > <i>Density: grams per cubic centimeter (g/cm³)</i> > <i>Conductivity: Siemens (S)</i> > <i>Acidity or alkalinity (pH)</i> <input type="checkbox"/> Energy: Joules (J or kJ) Determine the uncertainty of digital and analog measuring tools <ul style="list-style-type: none"> > <i>Usually ± 1 of the smallest division</i> <input type="checkbox"/> Calculate the % error of a given or recorded set of values <ul style="list-style-type: none"> > <i>$(\text{Measured} - \text{True}) / \text{True} \times 100$</i> <input type="checkbox"/> State values to appropriate numbers of significant figures <input type="checkbox"/> Calculate values to appropriate numbers of significant figures |
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Related Concepts:

- 01.1 Element symbols (Quia quiz: <http://www.quia.com/quiz/3246640.htm>)
- 01.2 Accuracy, Precision, Error, Uncertainty (Quia quiz: <http://www.quia.com/quiz/3507626.htm>)
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Command Terms: Be sure you understand what is being asked of you

Analyse Break down to identify key parts. Interpret information to reach conclusions.

<p>Apply</p> <p>Calculate</p> <p>Define</p> <p>Describe</p> <p>Design</p> <p>Determine</p> <p>Discuss</p> <p>Distinguish</p> <p>Evaluate</p> <p>Explain</p> <p>Identify</p> <p>Label</p> <p>List</p> <p>Outline</p> <p>Recall</p> <p>State</p> <p>Suggest/Propose</p>	<p>Use knowledge & understanding in response to a real situation or given circumstances.</p> <p>Obtain a numerical answer, showing relevant stages in the working.</p> <p>Give the precise meaning of a word, phrase, concept or physical quantity.</p> <p>Give a detailed account or picture of an event, pattern, process or pattern.</p> <p>Produce a plan, simulation or model.</p> <p>Obtain the only possible answer.</p> <p>Offer a considered and balanced review that includes a range of arguments, factors or hypotheses.</p> <p>Opinions or conclusions should be presented clearly and supported by appropriate evidence.</p> <p>Make clear the differences between two or more concepts or items.</p> <p>Assess the implications and limitations; make judgments about the ideas, works, solutions or ideas in relation to selected criteria.</p> <p>Give a detailed account including reasons or causes.</p> <p>Provide an answers from a number of possibilities. Recognise and state briefly a distinguishing fact or feature.</p> <p>Add title, labels or brief explanation(s) to a diagram or graph.</p> <p>Give a sequence of answers with no explanation.</p> <p>Give a brief account.</p> <p>Remember or recognize from prior learning experiences.</p> <p>Give a specific name, value or other brief answer without explanation or calculation.</p> <p>Propose a solution, hypothesis or other possible answer.</p>
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Introduction To Chemistry Lab Manual Answers

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Reviewing **Introduction To Chemistry Lab Manual Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Introduction To Chemistry Lab Manual Answers**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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