

# Gas Variables

How are the variables that describe a gas related?

## Why?

Imagine buying a balloon bouquet at a party store. How will the helium gas in the bouquet behave if you carry it outside on a hot summer day? How will it behave if you carry it outside during a snowstorm? What happens if the balloons are made of latex, which can stretch? What happens if the balloons are made of Mylar<sup>®</sup>, which cannot stretch? What if you add just a small amount of gas to each balloon? What if you add a lot of gas? In this activity, you will explore four variables that quantify gases—pressure (P), volume (V), temperature (T), and moles (n) of gas. These four variables can be related mathematically so that predictions about gas behavior can be made.

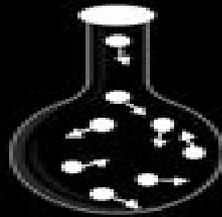
## Model 1 – Gases in a Nonflexible Container

### Experiment A (Adding more gas)



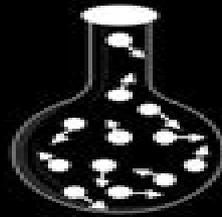
A1

Volume = 1 unit  
External pressure = 1 atm  
Internal pressure = 1 atm  
Temperature = 200 K



A2

Volume = 1 unit  
External pressure = 1 atm  
Internal pressure = 2 atm  
Temperature = 200 K



A3

Volume = 1 unit  
External pressure = 1 atm  
Internal pressure = 3 atm  
Temperature = 200 K

### Experiment B (Heating the gas)



B1

Volume = 1 unit  
External pressure = 1 atm  
Internal pressure = 1 atm  
Temperature = 200 K



B2

Volume = 1 unit  
External pressure = 1 atm  
Internal pressure = 2 atm  
Temperature = 400 K



B3

Volume = 1 unit  
External pressure = 1 atm  
Internal pressure = 3 atm  
Temperature = 600 K

\*Note: Volume in this model is recorded in units rather than liters because 4 molecules of gas at the conditions given would occupy a very small space ( $\sim 1 \times 10^{-22}$   $\mu\text{L}$ ). The particles shown here are much larger compared to the space between them than actual gas particles.

# Gas Variables Answers

**LP Steffe**



## Gas Variables Answers:

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**Hypersonic and High Temperature Gas Dynamics** John David Anderson, 1989 This book is a self contained text for those students and readers interested in learning hypersonic flow and high temperature gas dynamics It assumes no prior familiarity with either subject on the part of the reader If you have never studied hypersonic and or high temperature gas dynamics before and if you have never worked extensively in the area then this book is for you On the other hand if you have worked and or are working in these areas and you want a cohesive presentation of the fundamentals a development of important theory and techniques a discussion of the salient results with emphasis on the physical aspects and a presentation of modern thinking in these areas then this book is also for you In other words this book is designed for two roles 1 as an effective classroom text that can be used with ease by the instructor and understood with ease by the student and 2 as a viable professional working tool for engineers scientists and managers who have any contact in their jobs with hypersonic and or high temperature flow

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*Holt Chemistry* Ralph Thomas Myers,2004     *Gas* ,1961     Gas Engine ,1902

## Gas Variables Answers Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Gas Variables Answers**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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