

## Chapter 7 - Membrane Structure and Function

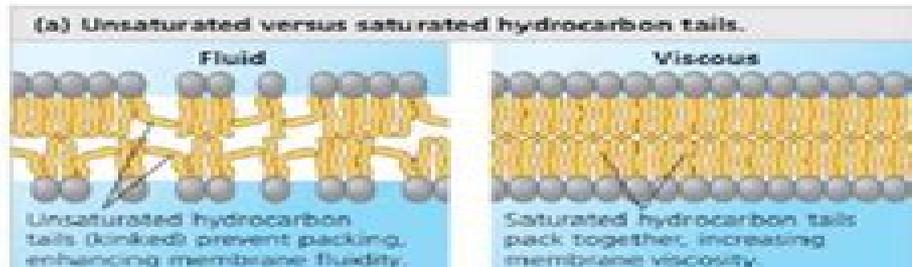
- Hydrophobic substances are nonpolar molecules that do not have affinity for water molecules
- Substances that contain polar bonds are hydrophilic
- Membrane phospholipids
  - The polar heads interact with water; the nonpolar tails do not
- Most prokaryotic cells have no internal membranes; eukaryotic cells do

### > Cellular membranes: a fluid mosaic of lipids and proteins

- Phospholipid is an amphipathic molecule: has both hydrophobic and hydrophilic regions
- Hydrophobic tails, hydrophilic heads
- Most membrane proteins are amphipathic
- Fluid mosaic model
  - The membrane is a mosaic of protein molecules bobbing in a fluid bilayer of phospholipids

### > fluidity of membranes

- A membrane is held together mainly by hydrophobic interactions (much weaker than covalent bonds)
- Most lipids can shift about sideways, and rarely also flip flop across the membrane, switching from one phospholipid layer to another
  - The sideways movement is rapid
- Some membrane proteins move in a highly directed manner, or driven by motor proteins
- A membrane remains fluid as temp decreases until the phospholipids settle into a closely packed arrangement and the membrane solidifies
  - Phospholipids with unsaturated hydrocarbon tails remain fluid for longer
  - Because of the kinks in the tails where double bonds are located, unsaturated hydrocarbon tails cannot pack together as closely, making the membrane more fluid



- Steroid cholesterol can affect fluidity
- At high temps, cholesterol makes membrane less fluid by restraining phospholipid movement
- Cholesterol is a "fluidity buffer", resisting changes in membrane fluidity that can be caused by changes in temp

# Membrane Structure And Function Chapter Notes

**Brendan G. Carr**



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