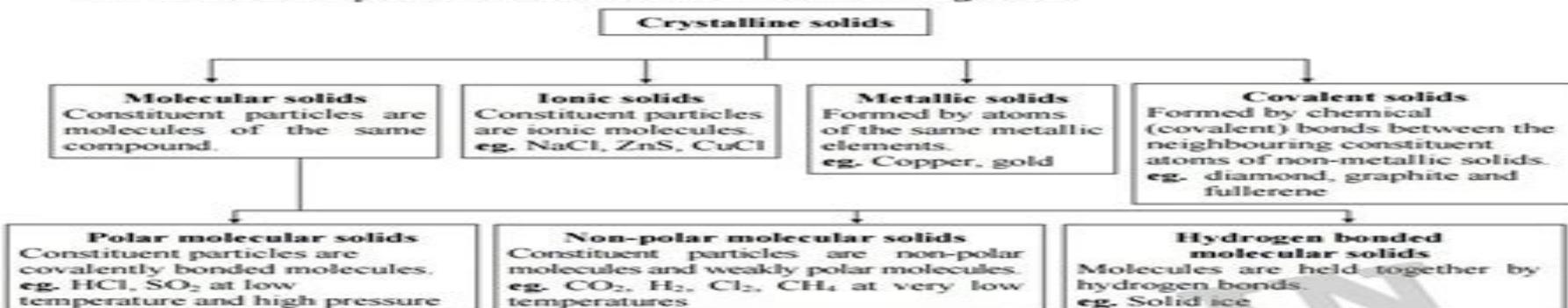


Chemistry - Classification & Formulas

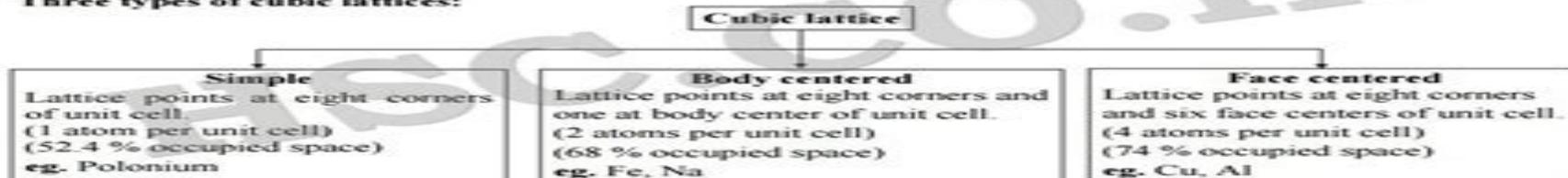
Classification of solids on the basis of the presence or absence of orderly arrangement of the constituent particles:



Classification of crystalline solids based on different binding forces:



Three types of cubic lattices:



Classification of solids based on response to magnetic field:

Substance	Characteristics	Magnetic alignment	Examples
Diamagnetic materials	<ul style="list-style-type: none"> Repelled weakly in magnetic field. All electrons are paired. 	$\downarrow\downarrow \downarrow\downarrow \downarrow\downarrow \downarrow\downarrow$	Benzene, NaCl, water, etc.
Paramagnetic materials	<ul style="list-style-type: none"> Weakly attracted in magnetic field. Unpaired electrons are present. Permanent magnetisation is not possible. 	$\swarrow \searrow \nearrow \nwarrow$	O ₂ , Cu ²⁺ , Fe ³⁺ , Cr ³⁺ , etc.
Ferromagnetic materials	<ul style="list-style-type: none"> Strongly attracted in magnetic field. Unpaired electrons are present. Permanent magnetisation is possible. Above the Curie temperature, ferromagnetic substance changes to paramagnetic substance. 	$\uparrow \uparrow \uparrow \uparrow \uparrow$	Fe, Ni, Co, CrO ₂ , etc.

Important Formulas

Density of unit cell:

$$\text{Mass of one atom} = \frac{\text{Atomic mass}}{\text{Avogadro number}}$$

ii. Volume of unit cell = a^3

$$\text{Density} = \frac{\text{Mass of unit cell}}{\text{Volume of unit cell}}$$

$$\text{Packing efficiency} = \frac{\text{Volume occupied by spheres in unit cell}}{\text{Volume of unit cell}} \times 100$$

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Siddappa N.Byrareddy



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