

Natural Sciences Grade 9
Marks: 35 converted to 20

Practical Task / Investigation
Memorandum

Acids, bases and the use of an indicator

PART 1

Table of results (For universal indicator)

Test tube	Colour	pH	Acid, base or neutral
1 (Vinegar)	Red	Any value below 7	Acid
2 (Water)	Green	7	Neutral
3 (Bicarbonate of soda)	Blue/Purple	Any value above 7	Base

Conclusion:

- (Distilled) water ✓ (1)
- 7 ✓ (1)
- Vinegar ✓ (1)
- The pH of an acid is (**below** ✓ / above) seven. (1)
- Bicarbonate of soda ✓ (1)
- The pH of a base is (below / **above** ✓) seven. (1)
- An indicator is a substance, which indicates whether a solution is an acid, a base or neutral ✓ through a colour change. ✓ (2)

PART 2

VERY IMPORTANT:

Teachers **MUST** do a test run or two themselves before the learners do this experiment. Results may vary a lot due to the different concentrations of the solutions.

Hypothesis:

When vinegar is treated with bicarbonate of soda ✓ the pH will increase. ✓

OR

When vinegar is treated with bicarbonate of soda ✓ the pH will decrease. ✓ (2)

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