

Part A: Concentration of Stock is 1.94×10^{-4} M
 Table 1. Preparation of Standard Solutions

Standard	Volume of Stock Solution (mL)	Final Volume of Diluted Solution (mL)	Concentration of Diluted Solutions (M)
1		50.00	1.94×10^{-4}
2		50.00	3.87×10^{-4}
3		50.00	9.70×10^{-4}
4		50.00	1.36×10^{-3}

Show your calculation for at least one value of volume of stock solution.

Part B:

Record the wavelength that had the highest absorbance for this dye: 525 nm

Part C:

Table 2. Absorbance data for standards and unknown.

Trial	Concentration (mol/L)	Absorbance (no unit)
1	1.94×10^{-4}	0.032
2	3.87×10^{-4}	0.084
3	9.70×10^{-4}	0.221
4	1.36×10^{-3}	0.268
Unknown Letter <u> </u>	(Answer from Q. #3, next page)	0.156
Equation of line:		

Analysis

Answer question 1 while working on the procedure for Part C (step 3):

1. Estimate the concentration of your unknown by comparing it to your standard solutions: (Circle one)

- | | |
|---------------------------------------|--------------------------------------|
| a. Less concentrated than standard #1 | d. Between standard #3 and #4 |
| b. Between standard #1 and #2 | e. More concentrated than standard # |
| c. Between standard #2 and #3 | |

Laboratory Investigation 30a Answers

Yi-Tong Ma



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