

General Chemistry Semester 2 Study Guide

Mr. Pitzer - Fall 2019

1st Period: Monday, June 17th, 8:00am - 10:00am

2nd Period: Wednesday, June 19th, 11:00am - 1:00pm

3rd Period: Thursday, June 20th, 11:00am - 1:00pm

Use this opportunity to review the course work. You may not yet be able to do this but please review all 3 weeks of material that we have done and discuss it. We will be having a final exam on the topics of equilibrium, kinetics, and acid-base chemistry. You will be expected to provide a full and complete answer to all questions. Please review your notes, lab, and journal for additional insight.

Please remember a passing grade on the semester depends on your performance on all quizzes. There is a full review and help session the week of midterms. It is highly recommended that you attend your study, test, and review session. Use the completed material to identify concepts you have not yet fully mastered and require additional practice.

On the final, students are allowed a calculator (graphing, programmable, scientific, and handheld).

Study session: Review Guide

Before your exam, the following resources may be available and that you may find helpful. You are encouraged to use them.

- 1. Periodic Table
- 2. Chemical Properties Table
- 3. Solubility Product Constants
- 4. Thermodynamic Tables
- 5. Review your notes on equilibrium, acids & bases, kinetics, thermodynamics
- 6. Lab experiments (focus those conducted in class)
- 7. ΔG° and ΔG
- 8. ΔH° and ΔS°
- 9. K and Q
- 10. Kinetics (initial, average, and instantaneous)

Final Exam: Reading Challenge

Review each week's text. Use following resources and information to assist in your reading. All of the following are available.

- 1. Molecular Structure of Gases (pp. 101 - 110)
 - a. 1. Molecular Structure of Gases (pp. 101 - 110)
 - b. Kinetic Theory
 - c. Molecular Structure
- 2. Kinetics (pp. 111 - 121, 124 - 131)
 - a. Rate Laws
 - b. Kinetic Theory
- 3. Thermodynamics (pp. 132 - 139)
 - a. Kinetic Theory
 - b. Specific Heat
- 4. Equilibrium (pp. 140 - 149)
 - a. Kinetics (pp. 111 - 121)
 - b. Rate Laws
 - c. Kinetic Theory
 - d. Kinetic Theory
- 5. Kinetics & Thermodynamics (pp. 140 - 149)
 - a. Kinetic Theory
 - b. Kinetic Theory
 - c. Kinetic Theory
 - d. Kinetic Theory
- 6. Kinetics & Thermodynamics (pp. 140 - 149)
 - a. Kinetic Theory
 - b. Kinetic Theory
 - c. Kinetic Theory
 - d. Kinetic Theory

General Chemistry Semester Two Study Guide

Ian Pickup



General Chemistry Semester Two Study Guide:

Decoding **General Chemistry Semester Two Study Guide**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**General Chemistry Semester Two Study Guide**," a mesmerizing literary creation penned with a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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