

Section 7 - Radioactivity
 BY HB SCIENCE RESOURCES

Activity 1: Alpha decay

Write down the equation for the alpha decay of ^{238}U .

$^{238}_{92}\text{U} \rightarrow ^{234}_{90}\text{Th} + ^4_2\text{He}$

Activity 2: Beta decay

Write down the equation for the beta decay of ^{14}C .

$^{14}_6\text{C} \rightarrow ^{14}_7\text{N} + e^- + \bar{\nu}_e$

Activity 3: Gamma decay

Write down the equation for the gamma decay of ^{60}Co .

$^{60}_{27}\text{Co} \rightarrow ^{60}_{27}\text{Co} + \gamma$

Activity 4: Half-life

Calculate the half-life of a substance if its activity decreases from 1000 Bq to 125 Bq in 10 years.

$t_{1/2} = 10 \text{ years}$

Activity 5: Radioactive dating

Explain how carbon-14 is used to date organic remains.

Activity 6: Nuclear fission

Write down the equation for the fission of ^{235}U .

$^{235}_{92}\text{U} + ^1_0\text{n} \rightarrow ^{141}_{54}\text{Xe} + ^{90}_{38}\text{Sr} + 2^1_0\text{n}$

Activity 7: Nuclear fusion

Write down the equation for the fusion of ^2_1H and ^3_1H .

$^2_1\text{H} + ^3_1\text{H} \rightarrow ^4_2\text{He} + ^1_0\text{n}$

Activity 8: Radioactive decay series

Describe the decay series of ^{238}U .

Activity 9: Radioactive decay and the environment

Explain the risks of radioactive contamination.

Activity 10: Radioactive decay and the future

Discuss the potential of nuclear energy.

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Holt Science Section 7 1 What Is Radioactivity Key

Christian G. Meyer



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