

Student Exploration: Half-Life

Prior Knowledge Questions (Do these *before* starting the Lesson.)

1. What do you think half-life means? (Does it mean you have until the point it is the halfway?)

Yes, I think it's halfway point.

2. If you had the opportunity to work with a half-life of a radioactive element, what is strange about it?

Radioactive elements that are used to date things that are old.

Warm-Up

Like a computer simulation, the simulation is **interactive** when you change its settings. It features some things to watch for: **variables** (what you can change) and **constants** (what you cannot change). In this process, what **variables** control the simulation? Is there any **constant** in the simulation?

The half-life of a radioactive element is a constant. The number of radioactive particles is a variable. The number of particles is constant and the half-life is a constant.

3. What do you think half-life is?

A constant property.

Notes: The following simulation uses three **variables** (initial amount, half-life, and time) and one **constant** (half-life). The variables are: initial amount, half-life, and time. The constant is: half-life.

4. What happens if the initial amount is 1000?

Nothing happens.

5. In the simulation, what is the **beginning** value of the process?

The value of the initial amount.

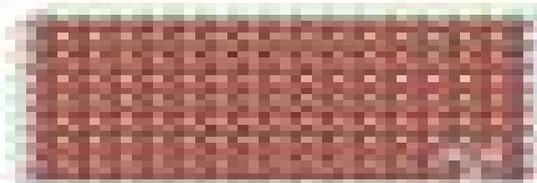


Figure 1: Simulation interface showing the initial state of the radioactive decay process.



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Puja Mehta



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