

PHYSICAL SCIENCES P.2



basic education

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Basic Education
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NATIONAL SENIOR CERTIFICATE

Diagram on page 9 is redrawn below for easy reference.



Changes to the conditions of equilibrium are made at different times. The graph shows the results obtained. X, Y and Z represent the gases in the above reaction.



4.5 Write down the TYPE of absorption in reaction II.

4.6 Compounds R and T are positional isomers.

The inorganic reagents shown below are available for reactions IV and V.

| | | | | |
|----|---------------------------------------|---------------------------|--------------|--------------|
| Bu | $\text{H}_2\text{SO}_4(\text{conc.})$ | $\text{NaOH}(\text{aq.})$ | HBr | H_2 |
|----|---------------------------------------|---------------------------|--------------|--------------|

4.7 Define for both chemical equilibrium.

Use the graph to answer the questions below.

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