

**I. Multiple Choice Questions (Type-I)**

1. On addition of conc.  $\text{H}_2\text{SO}_4$  to a chloride salt, colourless fumes are evolved but in case of an iodide salt, violet fumes come out. This is because

- (i)  $\text{H}_2\text{SO}_4$  reduces  $\text{HI}$  to  $\text{I}_2$
- (ii)  $\text{HI}$  is of violet colour
- (iii)  $\text{HI}$  gets oxidised to  $\text{I}_2$
- (iv)  $\text{HI}$  changes to  $\text{HIO}_3$

**Solution:**

Option (iii) is the answer.

2. In qualitative analysis when  $\text{H}_2\text{S}$  is passed through an aqueous solution of salt acidified with dil.  $\text{HCl}$ , a black precipitate is obtained. On boiling the precipitate with dil.  $\text{HNO}_3$ , it forms a solution of blue colour. Addition of excess of aqueous solution of ammonia to this solution gives \_\_\_\_\_.

- (i) a deep blue precipitate of  $\text{Cu}(\text{OH})_2$
- (ii) a deep blue solution of  $[\text{Cu}(\text{NH}_3)_4]^{2+}$
- (iii) a deep blue solution of  $\text{Cu}(\text{NO}_3)_2$
- (iv) a deep blue solution of  $\text{Cu}(\text{OH})_2 \cdot \text{Cu}(\text{NO}_3)_2$

**Solution:**

Option (ii) is the answer.

3. In a cyclotrimetaphosphoric acid molecule, how many single and double bonds are present?

- (i) 3 double bonds; 9 single bonds
- (ii) 6 double bonds; 6 single bonds
- (iii) 3 double bonds; 12 single bonds
- (iv) Zero double bonds; 12 single bonds

**Solution:**

Option (iii) is the answer.

4. Which of the following elements can be involved in  $\text{p}\pi\text{-d}\pi$  bonding?

- (i) Carbon
- (ii) Nitrogen
- (iii) Phosphorus
- (iv) Boron

**Solution:**

Option (iii) is the answer.

5. Which of the following pairs of ions are isoelectronic and isostructural?

- (i)  $\text{CO}_3^{2-}$ ,  $\text{NO}_3^-$
- (ii)  $\text{ClO}^-$ ,  $\text{CO}_3^{2-}$
- (iii)  $\text{SO}_3^{2-}$ ,  $\text{NO}_3^-$
- (iv)  $\text{ClO}^-$ ,  $\text{SO}_3^{2-}$

**Solution:**

Option (i) is the answer.

# Grade Chemistry Exemplar 2012

**SA Adler**



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