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Lei Huang



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so calculating enthalpy change of reaction in 2 $\Delta H_{\text{reaction}} = 6 \times \Delta H_{\text{f}}^{\circ}(\text{C}_2\text{H}_4) - 6 \times \Delta H_{\text{f}}^{\circ}(\text{C}_2\text{H}_2) - 6 \times \Delta H_{\text{f}}^{\circ}(\text{C}_2\text{H}_6) = 82 \text{ kJ}$

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