

Name : \_\_\_\_\_ Date : \_\_\_\_\_

# Lab Periodic Trends Worksheet

The atomic and the ionic radius of the first 18 elements of the periodic table are given below.

Element	H	He	Li	Be	B	C	N	O	F
Atomic Number	1	2	3	4	5	6	7	8	9
Atomic Radius (Å)	0.79	0.49	2.05	1.4	1.17	0.91	0.75	0.65	0.57
Ionic Radius (Å)	0.012	-	0.76	0.35	0.23	0.16	0.13	1.4	1.33

Ne	Na	Mg	Al	Si	P	S	Cl	Ar
10	11	12	13	14	15	16	17	18
0.38	2.23	1.72	1.82	1.46	1.23	1.09	0.97	0.71
-	1.02	0.72	0.535	0.4	0.38	0.37	1.81	-

Using the data above, construct the line graph of atomic radius vs. atomic number and ionic radius vs. the atomic number.

**Atomic Radius vs. Atomic Number**



**Ionic Radius vs. Atomic Number**



# Exploring Periodic Trends Lab Answer Key

**JA Banks**



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## **Table of Contents Exploring Periodic Trends Lab Answer Key**

1. Understanding the eBook Exploring Periodic Trends Lab Answer Key
  - The Rise of Digital Reading Exploring Periodic Trends Lab Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Exploring Periodic Trends Lab Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Exploring Periodic Trends Lab Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Exploring Periodic Trends Lab Answer Key
  - Personalized Recommendations
  - Exploring Periodic Trends Lab Answer Key User Reviews and Ratings
  - Exploring Periodic Trends Lab Answer Key and Bestseller Lists
5. Accessing Exploring Periodic Trends Lab Answer Key Free and Paid eBooks
  - Exploring Periodic Trends Lab Answer Key Public Domain eBooks
  - Exploring Periodic Trends Lab Answer Key eBook Subscription Services
  - Exploring Periodic Trends Lab Answer Key Budget-Friendly Options

6. Navigating Exploring Periodic Trends Lab Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Exploring Periodic Trends Lab Answer Key Compatibility with Devices
  - Exploring Periodic Trends Lab Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Exploring Periodic Trends Lab Answer Key
  - Highlighting and Note-Taking Exploring Periodic Trends Lab Answer Key
  - Interactive Elements Exploring Periodic Trends Lab Answer Key
8. Staying Engaged with Exploring Periodic Trends Lab Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Exploring Periodic Trends Lab Answer Key
9. Balancing eBooks and Physical Books Exploring Periodic Trends Lab Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Exploring Periodic Trends Lab Answer Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Exploring Periodic Trends Lab Answer Key
  - Setting Reading Goals Exploring Periodic Trends Lab Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Exploring Periodic Trends Lab Answer Key
  - Fact-Checking eBook Content of Exploring Periodic Trends Lab Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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