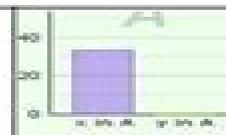


**Activity A:**  
**Temperature and diffusion**

Get the Gizmo ready:

- Click **Reset** (↺).
- Set the **Wall** to 100%.



**Introduction:** In this Gizmo, temperature is measured on the **Kelvin scale**. On this scale, 0 K represents **absolute zero**, the coldest possible temperature. Water freezes at 273.15 K (0 °C), and water boils at 373.15 K (100 °C).

**Question:** How does temperature affect the rate of diffusion?

1. **Observe:** Click **Reset**. Set the temperature to 200 K, and click **Play**. Observe the number of particles that pass through the wall. Then set the temperature to 300 K, and **Play** again.
- How does the temperature of the gas relate to the number of particles that pass through the wall?

As the temperature of the gas increases, the number of particles that pass through the wall increases.

The number of particles that pass through the wall is a measure of the average **kinetic energy** of the particles. Kinetic energy (KE) depends on the velocity and mass of the particles.  $KE = \frac{1}{2}mv^2$ .

2. **Observe:** How do you think temperature will affect the rate of diffusion?

As the temperature increases, the rate of diffusion will also increase.

3. **Observe:** Click **Reset**. Set the wall to 50% and click **Play**. Set the temperature to 200 K. Observe the number of particles that pass through the wall. Then click **Play** again.

How does the rate of diffusion change as the temperature increases? How do you think this is related to the kinetic energy of the particles?

Report the number of particles that pass through the wall at 200 K and 300 K.

Temp = 200 K: \_\_\_\_\_

Temp	Number of particles that pass through the wall
200 K	_____
300 K	_____

# Explore Learning Gizmo Answer Key Diffusion

**G Psacharopoulos**



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