

## Half Life Worksheet – Extra Practice

- 1) Fluorine-21 has a half life of approximately 5 seconds. What fraction of the original nuclei would remain after 1 minute?  $\frac{1}{4096}$
- 2) Iodine-131 has a half life of 8 days. What fraction of the original sample would remain at the end of 32 days?  $\frac{1}{64}$
- ★ 3) The half-life of chromium-51 is 28 days. If the sample contained 510 grams, how much chromium would remain after 56 days? How much would remain after 1 year? How much was present 168 days ago?  $\rightarrow 12$   
 $\rightarrow .0629$   $\rightarrow 32 \text{ g}$
- ★ 4) If 20.0 g of a radioactive isotope are present at 1:00 PM and 5.0 g remain at 2:00 PM, what is the half life of the isotope? 30 minutes
- 5) The half life of Uranium-238 is 4.5 billion years and the age of earth is  $4.5 \times 10^9$  years. What fraction of Uranium-238 that was present when Earth was formed still remains?  $\frac{1}{2}$
- 6) Chromium-48 decays. After 6 half-lives, what fraction of the original nuclei would remain?  $\frac{1}{64}$
- 7) The half life of iodine-125 is 60 days. What fraction of iodine-125 nuclides would be left after 360 days?  $\frac{1}{64}$
- 8) Titanium-51 decays with a half life of 6 minutes. What fraction of titanium would remain after one hour?  $\frac{1}{1024}$
- 9) A medical institution requests 1 g of bismuth-214, which has a half life of 20 min. How many grams of bismuth-214 must be prepared if the shipping time is 2 h? 64 g
- ★ 10) The half life of radium 226 is 1602 years. If you have 500 grams of radium today how many grams would have been present 9612 years ago? 32,000g

★ = more like what will be on test

# Explore Learning Half Life Wkst Answer Key

**CL Gary**



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## **Table of Contents Explore Learning Half Life Wkst Answer Key**

1. Understanding the eBook Explore Learning Half Life Wkst Answer Key
  - The Rise of Digital Reading Explore Learning Half Life Wkst Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Explore Learning Half Life Wkst Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Explore Learning Half Life Wkst Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Explore Learning Half Life Wkst Answer Key
  - Personalized Recommendations
  - Explore Learning Half Life Wkst Answer Key User Reviews and Ratings
  - Explore Learning Half Life Wkst Answer Key and Bestseller Lists
5. Accessing Explore Learning Half Life Wkst Answer Key Free and Paid eBooks
  - Explore Learning Half Life Wkst Answer Key Public Domain eBooks
  - Explore Learning Half Life Wkst Answer Key eBook Subscription Services
  - Explore Learning Half Life Wkst Answer Key Budget-Friendly Options
6. Navigating Explore Learning Half Life Wkst Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Explore Learning Half Life Wkst Answer Key Compatibility with Devices
  - Explore Learning Half Life Wkst Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Explore Learning Half Life Wkst Answer Key
  - Highlighting and Note-Taking Explore Learning Half Life Wkst Answer Key
  - Interactive Elements Explore Learning Half Life Wkst Answer Key

8. Staying Engaged with Explore Learning Half Life Wkst Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Explore Learning Half Life Wkst Answer Key
9. Balancing eBooks and Physical Books Explore Learning Half Life Wkst Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Explore Learning Half Life Wkst Answer Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Explore Learning Half Life Wkst Answer Key
  - Setting Reading Goals Explore Learning Half Life Wkst Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Explore Learning Half Life Wkst Answer Key
  - Fact-Checking eBook Content of Explore Learning Half Life Wkst Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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