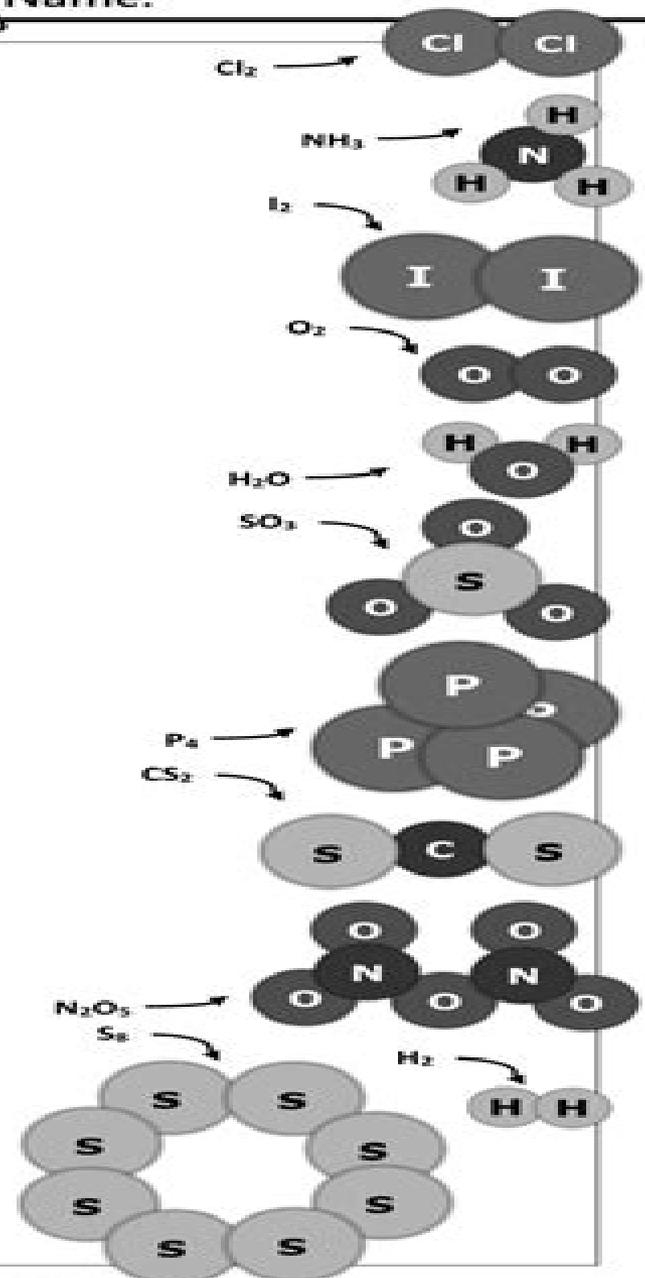


Balance the following chemical equations.

1. $\text{___ Na} + \text{___ Cl}_2 \rightarrow \text{___ NaCl}$
2. $\text{___ Ca} + \text{___ O}_2 \rightarrow \text{___ CaO}$
3. $\text{___ Na} + \text{___ O}_2 \rightarrow \text{___ Na}_2\text{O}$
4. $\text{___ N}_2 + \text{___ H}_2 \rightarrow \text{___ NH}_3$
5. $\text{___ K} + \text{___ I}_2 \rightarrow \text{___ KI}$
6. $\text{___ Al} + \text{___ O}_2 \rightarrow \text{___ Al}_2\text{O}_3$
7. $\text{___ H}_2 + \text{___ O}_2 \rightarrow \text{___ H}_2\text{O}$
8. $\text{___ H}_2\text{O}_2 \rightarrow \text{___ H}_2\text{O} + \text{___ O}_2$
9. $\text{___ Li} + \text{___ Cl}_2 \rightarrow \text{___ LiCl}$
10. $\text{___ SO}_2 + \text{___ O}_2 \rightarrow \text{___ SO}_3$
11. $\text{___ S}_8 + \text{___ O}_2 \rightarrow \text{___ SO}_3$
12. $\text{___ P} + \text{___ O}_2 \rightarrow \text{___ P}_2\text{O}_5$
13. $\text{___ P}_4 + \text{___ O}_2 \rightarrow \text{___ P}_2\text{O}_5$
14. $\text{___ N}_2 + \text{___ F}_2 \rightarrow \text{___ NF}_3$
15. $\text{___ S}_8 + \text{___ C} \rightarrow \text{___ CS}_2$
16. $\text{___ N}_2 + \text{___ O}_2 \rightarrow \text{___ N}_2\text{O}_5$
17. $\text{___ Mg} + \text{___ Cl}_2 \rightarrow \text{___ MgCl}_2$
18. $\text{___ Rb} + \text{___ P} \rightarrow \text{___ Rb}_3\text{P}$
19. $\text{___ Fe} + \text{___ O}_2 \rightarrow \text{___ Fe}_2\text{O}_3$
20. $\text{___ K} + \text{___ Br}_2 \rightarrow \text{___ KBr}$
21. $\text{___ HgO} \rightarrow \text{___ O}_2 + \text{___ Hg}$
22. $\text{___ H}_2 + \text{___ Cl}_2 \rightarrow \text{___ HCl}$
23. $\text{___ Zn} + \text{___ S}_8 \rightarrow \text{___ ZnS}$
24. $\text{___ Ag}_2\text{O} \rightarrow \text{___ O}_2 + \text{___ Ag}$



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Ying-Ying Zheng



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