

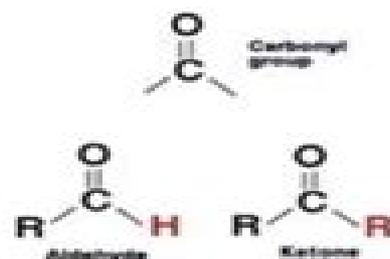
Identification of Unknown Organic Compound

IN THIS EXPERIMENT, YOU WILL

- Determine the classes of alcohol by using the Lucas Test
- Identify the carbonyl compound by using Brady's, Tollen's and Fehling's Test.
- Identify the methyl carbonyl and methyl carbinol compounds by using the Iodoform Test

INTRODUCTION

Alcohols are organic compounds in which, as their functional group, one or more hydrogen atoms in an alkane have been replaced by a group of $-OH$ (hydroxyl). Alcohols fall into various classes depending on how the $-OH$ group is positioned on the carbon atom chain. Alcohol can be oxidized to aldehyde, ketone, and carboxylic acid. The product formed from the oxidation depends on the classes of alcohol used as reactant.



Aldehydes and ketones are carbonyl compounds. They contained a carbonyl group; a double bond of carbon-oxygen, (C=O). Oxygen is far more electronegative than carbon, and thus has a strong tendency to pull electrons towards itself in a carbon - oxygen bond. That makes the bond between carbon and oxygen extremely polar. Nucleophiles can attack the partially positive carbon-atom in the carbonyl group. Aldehydes have the $RCHO$ structure, in which R is an alkyl or aryl group. The hydrogen atom attached to the carbonyl group leads to different oxidation behavior in aldehydes compared to that seen with ketones. Ketones meanwhile have the $RR'CO$ structure where R and R' are an alkyl group attached to both ends of carbon in a carbonyl group.

Methods To Identify Unknown Compounds

Michael Brown



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Table of Contents Methods To Identify Unknown Compounds

1. Understanding the eBook Methods To Identify Unknown Compounds
 - The Rise of Digital Reading Methods To Identify Unknown Compounds
 - Advantages of eBooks Over Traditional Books
2. Identifying Methods To Identify Unknown Compounds
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Methods To Identify Unknown Compounds
 - User-Friendly Interface
4. Exploring eBook Recommendations from Methods To Identify Unknown Compounds
 - Personalized Recommendations
 - Methods To Identify Unknown Compounds User Reviews and Ratings
 - Methods To Identify Unknown Compounds and Bestseller Lists
5. Accessing Methods To Identify Unknown Compounds Free and Paid eBooks

- Methods To Identify Unknown Compounds Public Domain eBooks
 - Methods To Identify Unknown Compounds eBook Subscription Services
 - Methods To Identify Unknown Compounds Budget-Friendly Options
6. Navigating Methods To Identify Unknown Compounds eBook Formats
 - ePub, PDF, MOBI, and More
 - Methods To Identify Unknown Compounds Compatibility with Devices
 - Methods To Identify Unknown Compounds Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Methods To Identify Unknown Compounds
 - Highlighting and Note-Taking Methods To Identify Unknown Compounds
 - Interactive Elements Methods To Identify Unknown Compounds
 8. Staying Engaged with Methods To Identify Unknown Compounds
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Methods To Identify Unknown Compounds
 9. Balancing eBooks and Physical Books Methods To Identify Unknown Compounds
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Methods To Identify Unknown Compounds
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Methods To Identify Unknown Compounds
 - Setting Reading Goals Methods To Identify Unknown Compounds
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Methods To Identify Unknown Compounds
 - Fact-Checking eBook Content of Methods To Identify Unknown Compounds
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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