

- The entropy change of the surroundings is driven by heat flow.
  - In total, every energy transfer or transformation increases the entropy of the universe.
- c. Conservation of matter in chemical systems.
- Matter neither created nor destroyed, so we must have the same number and type of atoms after the chemical change as were present before the chemical change.
- d. Kinetic and potential energy
- **Kinetic energy:** energy in motion
  - **Potential energy:** the energy possessed by a body by virtue of its position relative to others, stresses within itself, electric charge, and other factors.
- e. Transformations between different forms of energy (thermal, chemical, radiant, nuclear, mechanical, electrical, electromagnetic, sound, and thermal).
- **Energy Transformations Examples:** An energy transformation is the change of energy from one form to another. Energy transformations occur everywhere every second of the day. There are many different forms of energy such as electrical, thermal, nuclear, mechanical, electromagnetic, sound, and thermal.

### f. Differences between chemical and physical properties/changes

- The difference between a physical reaction and a chemical reaction is composition.
- **Chemical property change:** The changing of color of a substance is not necessarily an indication of a chemical change. For example, changing the color of a metal does not change its physical properties however, in a chemical reaction, a color change is usually an indicator that a reaction is occurring.
  - Change in Temperature
  - Change in Color
  - Noticeable Odor (after reaction has begun)
  - Formation of a Precipitate
  - Formation of Bubbles
- **Physical property change:** In a physical change there is a difference in the appearance, smell, or simple display of a sample of matter without a change in composition.
  - Texture
  - Color
  - Temperature
  - Shape
  - Change of Boiling Point and Melting Point are significant factors in determining this change.

Physical properties include many other aspects of a substance. The following are (but not limited to) physical properties:

- Luster
- Malleability
- Ability to be drawn into a thin wire
- Density
- Volatility
- Solubility
- Mass
- Volume

### g. Various temperature scales (Celsius, Fahrenheit, Kelvin)

# General Science Praxis Study Guide

**Jin-Ying Zhang**



## **General Science Praxis Study Guide:**

## The Enigmatic Realm of **General Science Praxis Study Guide**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **General Science Praxis Study Guide** a literary masterpiece penned by a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of those who partake in its reading experience.

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