

# Gases | Study Guide

## Test Format

Multiple Choice  
Gas Law Problems  
Gas Stoich Problems  
Short Answer

## Equations & Conversions

$$P = \frac{F}{A}$$

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

$$PV = nRT$$

$$R = 0.0821 \text{ L}\cdot\text{atm}/\text{mol}\cdot\text{K}$$

$$1 \text{ atm} = 760 \text{ mmHg} = 101.3 \text{ kPa}$$

$$\frac{\text{Rate of Effusion of L}}{\text{Rate of Effusion of H}} = \sqrt{\frac{\text{Molar Mass of H}}{\text{Molar Mass of L}}}$$

$$K = ^\circ\text{C} + 273 \quad 0.001 \text{ kL} = 1 \text{ L} = 1,000 \text{ mL}$$

$$\text{At STP, 1 mole} = 22.4 \text{ L} \quad \text{KHDSdcm}$$

1. A box weighs 180 lbs and has a base that is 6 inches by 9 inches. What is the pressure this box exerts on the surface it is sitting on?

$$P = \frac{F}{A} = \frac{180 \text{ lbs}}{54 \text{ in}^2} = \boxed{3.3 \text{ psi}}$$

$$54 \text{ in}^2 = \text{area}$$

2. What is atmospheric pressure?

The pressure exerted by the collisions of gases in the air surrounding Earth ( $\text{N}_2$ ,  $\text{O}_2$ , etc).

3. Convert 1.088 atm to:

a. mm Hg  $\frac{1.088 \text{ atm} \cdot 760 \text{ mmHg}}{1 \text{ atm}} = \boxed{826.9 \text{ mmHg}}$

b. kPa  $\frac{1.088 \text{ atm} \cdot 101.3 \text{ kPa}}{1 \text{ atm}} = \boxed{110.2 \text{ kPa}}$

4. What is a barometer and who invented it?

A tool that measures room/air pressure. Invented by Torricelli.



5. How are temperature and kinetic energy related?

Direct. As temp  $\uparrow$ , KE  $\uparrow$ , molecules move faster.  
As temp  $\downarrow$ , KE  $\downarrow$ , molecules move slower.

6. Briefly describe the five assumptions of the kinetic-molecular theory.

Gases have a lot of tiny particles that are always in motion that is random, fast, and continuous. Particles are far apart without any forces of attraction, so their collisions are elastic  $\rightarrow$  bouncing off of each other/walls. The speed of the molecules change with temperature.

# Gases Unit Study Guide Answers

**Clemens Wendtner**



**Gases Unit Study Guide Answers:**

This book delves into Gases Unit Study Guide Answers. Gases Unit Study Guide Answers is a vital topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Gases Unit Study Guide Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
    - Chapter 1: Introduction to Gases Unit Study Guide Answers
    - Chapter 2: Essential Elements of Gases Unit Study Guide Answers
    - Chapter 3: Gases Unit Study Guide Answers in Everyday Life
    - Chapter 4: Gases Unit Study Guide Answers in Specific Contexts
    - Chapter 5: Conclusion
  2. In chapter 1, the author will provide an overview of Gases Unit Study Guide Answers. The first chapter will explore what Gases Unit Study Guide Answers is, why Gases Unit Study Guide Answers is vital, and how to effectively learn about Gases Unit Study Guide Answers.
  3. In chapter 2, the author will delve into the foundational concepts of Gases Unit Study Guide Answers. The second chapter will elucidate the essential principles that need to be understood to grasp Gases Unit Study Guide Answers in its entirety.
  4. In chapter 3, this book will examine the practical applications of Gases Unit Study Guide Answers in daily life. This chapter will showcase real-world examples of how Gases Unit Study Guide Answers can be effectively utilized in everyday scenarios.
  5. In chapter 4, this book will scrutinize the relevance of Gases Unit Study Guide Answers in specific contexts. The fourth chapter will explore how Gases Unit Study Guide Answers is applied in specialized fields, such as education, business, and technology.
  6. In chapter 5, the author will draw a conclusion about Gases Unit Study Guide Answers. The final chapter will summarize the key points that have been discussed throughout the book.
- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Gases Unit Study Guide Answers.

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