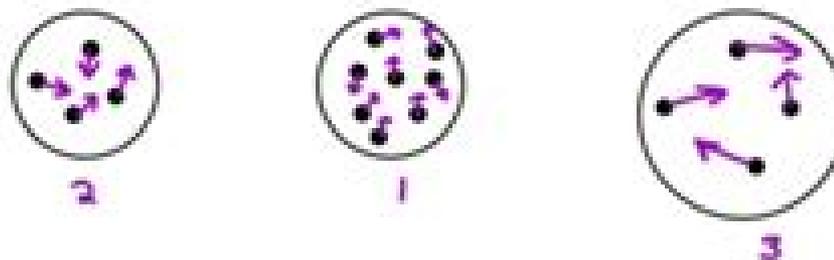


20. For each experiment in Model 2, determine the relationship between the independent and dependent variables, and write an algebraic expression for the relationship using variables that relate to those in the experiment ( $P$ ,  $V$ ,  $T$  or  $n$ ). Use  $k$  as a proportionality constant in each equation.

Constant Pressure

|                               | Experiment C  | Experiment D  | Experiment E      |
|-------------------------------|---------------|---------------|-------------------|
| Direct or Inverse Proportion? | <i>Direct</i> | <i>Direct</i> | <i>inverse</i>    |
| Algebraic Expression          | $4V = n$      | $200V = T$    | $\frac{1}{V} = P$ |

21. The three samples of identical gas molecules below all have the same internal pressure. Rank the samples from lowest temperature to highest temperature, and add arrows of appropriate size to illustrate the average kinetic energy of the molecules in the samples.



# Gas Variable Model Answer Key

**Tobias Bleicker**



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