

### Polyatomic Ion Worksheet

Place the charges on each ion – Write the Formula - Name the Compound

Remember – The cation always goes first.

Example:	$\text{NH}_4^{+1}$	&	$\text{I}^{-1}$	Formula:	$\text{NH}_4\text{I}$	Name:	<u>Ammonium Iodide</u>
1.	$\text{NH}_4$	&	$\text{O}$	Formula:	_____	Name:	_____
2.	$\text{Ca}$	&	$\text{CO}_3$	Formula:	_____	Name:	_____
3.	$\text{HCO}_3$	&	$\text{Na}$	Formula:	_____	Name:	_____
4.	$\text{NO}_3$	&	$\text{Mg}$	Formula:	_____	Name:	_____
5.	$\text{K}$	&	$\text{SO}_4$	Formula:	_____	Name:	_____
6.	$\text{Li}$	&	$\text{PO}_4$	Formula:	_____	Name:	_____

### Finding the Formula from the Name

Example:	Lithium Chloride	Ions:	$\text{Li}^{+1}$ & $\text{Cl}^{-1}$	Formula:	$\text{LiCl}$
1.	Sodium Oxide	Ions:	_____	Formula:	_____
2.	Potassium Bromide	Ions:	_____	Formula:	_____
3.	Calcium Nitride	Ions:	_____	Formula:	_____
4.	Beryllium Sulfide	Ions:	_____	Formula:	_____
5.	Magnesium Iodide	Ions:	_____	Formula:	_____
6.	Aluminum Oxide	Ions:	_____	Formula:	_____
7.	Ammonium Bromide	Ions:	_____	Formula:	_____
8.	Calcium Bicarbonate	Ions:	_____	Formula:	_____
9.	Potassium Phosphate	Ions:	_____	Formula:	_____
10.	Aluminum Sulfate	Ions:	_____	Formula:	_____

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**Wolfgang Guggemos**



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## **Table of Contents Formulas With Polyatomic Ions Worksheet Page 57**

1. Understanding the eBook Formulas With Polyatomic Ions Worksheet Page 57
  - The Rise of Digital Reading Formulas With Polyatomic Ions Worksheet Page 57
  - Advantages of eBooks Over Traditional Books
2. Identifying Formulas With Polyatomic Ions Worksheet Page 57
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Formulas With Polyatomic Ions Worksheet Page 57
  - User-Friendly Interface
4. Exploring eBook Recommendations from Formulas With Polyatomic Ions Worksheet Page 57

- Personalized Recommendations
  - Formulas With Polyatomic Ions Worksheet Page 57 User Reviews and Ratings
  - Formulas With Polyatomic Ions Worksheet Page 57 and Bestseller Lists
5. Accessing Formulas With Polyatomic Ions Worksheet Page 57 Free and Paid eBooks
    - Formulas With Polyatomic Ions Worksheet Page 57 Public Domain eBooks
    - Formulas With Polyatomic Ions Worksheet Page 57 eBook Subscription Services
    - Formulas With Polyatomic Ions Worksheet Page 57 Budget-Friendly Options
  6. Navigating Formulas With Polyatomic Ions Worksheet Page 57 eBook Formats
    - ePub, PDF, MOBI, and More
    - Formulas With Polyatomic Ions Worksheet Page 57 Compatibility with Devices
    - Formulas With Polyatomic Ions Worksheet Page 57 Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Formulas With Polyatomic Ions Worksheet Page 57
    - Highlighting and Note-Taking Formulas With Polyatomic Ions Worksheet Page 57
    - Interactive Elements Formulas With Polyatomic Ions Worksheet Page 57
  8. Staying Engaged with Formulas With Polyatomic Ions Worksheet Page 57
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Formulas With Polyatomic Ions Worksheet Page 57
  9. Balancing eBooks and Physical Books Formulas With Polyatomic Ions Worksheet Page 57
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Formulas With Polyatomic Ions Worksheet Page 57
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Formulas With Polyatomic Ions Worksheet Page 57
    - Setting Reading Goals Formulas With Polyatomic Ions Worksheet Page 57
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Formulas With Polyatomic Ions Worksheet Page 57

- Fact-Checking eBook Content of Formulas With Polyatomic Ions Worksheet Page 57
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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