

QUESTION – CHEMISTRY

FORM -2

CHEMICAL FAMILIES

1. Study the information in the table below and answer the questions that follow:

Element	Atomic radius (nm)	Ionic radius (nm)
W	0.114	0.195
X	0.072	0.136
Y	0.133	0.216
Z	0.099	0.181

(a) Would these form part of a metallic or a non-metallic group? Explain

(b) Suggest an element in the table above likely to be the most reactive. Explain

2. State the reason for using Argon in electric light bulbs

3. Study the information in the table below and answer the questions that follow. The letters do not represent the actual symbols of the elements.

Element	Electronic configuration	Boiling point
X	2.7	-188°C
Y	2.8.7	-35°C
Z	2.8.8.7	59°C

(a) What is the general name given to the group in which the elements X, Y and Z belong?

Form Chemistry Questions

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Adrefiza, Aodah Diamah, Ismarita
Ramayanti**

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Table of Contents Form Chemistry Questions

1. Understanding the eBook Form Chemistry Questions
 - The Rise of Digital Reading Form Chemistry Questions
 - Advantages of eBooks Over Traditional Books
2. Identifying Form Chemistry Questions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Form Chemistry Questions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Form Chemistry Questions
 - Personalized Recommendations
 - Form Chemistry Questions User Reviews and Ratings
 - Form Chemistry Questions and Bestseller Lists

5. Accessing Form Chemistry Questions Free and Paid eBooks
 - Form Chemistry Questions Public Domain eBooks
 - Form Chemistry Questions eBook Subscription Services
 - Form Chemistry Questions Budget-Friendly Options
6. Navigating Form Chemistry Questions eBook Formats
 - ePub, PDF, MOBI, and More
 - Form Chemistry Questions Compatibility with Devices
 - Form Chemistry Questions Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Form Chemistry Questions
 - Highlighting and Note-Taking Form Chemistry Questions
 - Interactive Elements Form Chemistry Questions
8. Staying Engaged with Form Chemistry Questions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Form Chemistry Questions
9. Balancing eBooks and Physical Books Form Chemistry Questions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Form Chemistry Questions
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Form Chemistry Questions
 - Setting Reading Goals Form Chemistry Questions
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Form Chemistry Questions
 - Fact-Checking eBook Content of Form Chemistry Questions
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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