

Limiting reactants and Theoretical Yield

·How to recognize when you have a Limiting Reactant problem: In the problem will be **given** the mass or moles of two (2) reactants.

·You must **decide** which reactant is used up first.

That is the **limiting reactant**. The limiting reactant gives the least **amount of Product** (Theoretical Yield).

·**The process:**

1) Calculate the **mass of product** possible from each reactant. Use the roadmap.

2) Compare the two results to see which gives the least **amount of product**. (theoretical yield)

Limiting Reactant And Limiting Yield Answer Key

G Psacharopoulos



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