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### Unit 6 - Chemical Bonding Test Review

1. The OCTET RULE states that atoms try to fill up their \_\_\_\_\_ and \_\_\_\_\_ orbitals in order to attain 8 valence electrons.
2. When drawing Lewis Dot structures, what do the dots around the symbol represent?
3. Draw Lewis Dot structures for the following compounds:



4. Fill in the T-Chart below by completing the definition of each type of bond and providing 2 examples for each.

	Ionic	Covalent
Definition:	A bond between elements from _____ sides of the periodic table; a bond between elements with very _____ electronegativities.	A bond between elements _____ each other on the periodic table; a bond between elements with _____ electronegativities.
Examples:		

# Modeling Chemistry Unit 6 Test Review

**Linda L. Murphy, Jane Close  
Conoley, James C. Impara**



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