

Name \_\_\_\_\_

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### Limiting Reactant Lab

**Pre-lab Questions:** Use the background information in the lesson and the video to help you answer these questions before the lab.

1. What is a limiting reactant? (10pts)

A limiting reactant in a chemical reaction determines how much product will be produced.

2. What is an excess reactant? (10pts)

An excess reactant in a chemical reaction is present in an amount greater than what will be used to complete the reaction

3. What is the purpose of this experiment? (10pts)

to see the reaction of aluminum foil when placed in an aqueous solution of copper (II) chloride to produce copper and aluminum chloride. The purpose is to determine which of the reactants will limit the production of copper

**Procedures:** Use the video in the lesson to help you complete the data table and post-lab questions below.

Data & Observation Table (12 points)	
Mass of Al (g)	0.12 g
Volume of CuCl <sub>2</sub> (mL)	50.9 mL
Color of CuCl <sub>2</sub> solution before reaction	green
Color of Al before reaction	Silver
Color of CuCl <sub>2</sub> solution after reaction	dark cloudy green
Color of Al after the reaction	silver and brown

4. Balance the equation below: (8 pts)



# Limiting Reactant Lab Report Answers

**Daniela Niemeyer**



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