

STUDY GUIDE FOR MODULE NO. 4

HYDROCARBON CHAINS AND RINGS



MODULE OVERVIEW

Many hydrocarbons occur in nature. In addition to making up fossil fuels, they are present in trees and plants, as, for example, in the form of pigments called carotenes that occur in carrots and green leaves. More than 98 percent of natural crude rubber is a hydrocarbon polymer, a chainlike molecule consisting of many units linked together. The structures and chemistry of individual hydrocarbons depend in large part on the types of chemical bonds that link together the atoms of their constituent molecules.

Nineteenth-century chemists classified hydrocarbons as either aliphatic or aromatic on the basis of their sources and properties. Aliphatic (from Greek *alipha*, "fat") described hydrocarbons derived by chemical degradation of fats or oils. Aromatic hydrocarbons constituted a group of related substances obtained by chemical degradation of certain pleasant-smelling plant extracts. The terms *aliphatic* and *aromatic* are retained in modern terminology, but the compounds they describe are distinguished on the basis of structure rather than origin.



MODULE LEARNING OBJECTIVES

At the end of the unit, the student should be able to:

1. Identify the structures and homologous members of alkene, alkane and alkyne families.
2. Describe the structure of benzene and the substituted aromatics.
3. Name the organic compounds using IUPAC.
4. Differentiate the types of structural compounds.



LESSON 1: HYDROCARBONS: REPRESENTATIVE ALKANES, ALKENES, ALKYNES, AND AROMATIC COMPOUNDS

HYDROCARBONS

The simplest organic compounds are those composed of only two elements: carbon and hydrogen. These compounds are called hydrocarbons. Hydrocarbons themselves are separated into two types: aliphatic hydrocarbons and aromatic hydrocarbons. Aliphatic hydrocarbons are hydrocarbons based on chains of C atoms. There are three types of aliphatic hydrocarbons. Alkanes are aliphatic hydrocarbons with only single covalent bonds. Alkenes are aliphatic hydrocarbons that contain at least one C=C double bond, and alkynes are aliphatic hydrocarbons that contain a C=C triple bond. Occasionally, we find an aliphatic hydrocarbon with a ring of C atoms; these hydrocarbons are called cycloalkanes (or cycloalkenes or cycloalkynes).

The general molecular formulas and molecular structures of the alkanes, alkenes, and alkynes.

Group	General Molecular Formula	Example Compound	Molecular Structure
Alkanes	C_nH_{2n+2}	Ethane	
Alkenes	C_nH_{2n}	Ethene	
Alkynes	C_nH_{2n-2}	Ethyne	

Hydrocarbons Study Guide

A Loxley



Hydrocarbons Study Guide:

The book delves into Hydrocarbons Study Guide. Hydrocarbons Study Guide is a crucial topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Hydrocarbons Study Guide, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Hydrocarbons Study Guide
 - Chapter 2: Essential Elements of Hydrocarbons Study Guide
 - Chapter 3: Hydrocarbons Study Guide in Everyday Life
 - Chapter 4: Hydrocarbons Study Guide in Specific Contexts
 - Chapter 5: Conclusion
2. In chapter 1, the author will provide an overview of Hydrocarbons Study Guide. This chapter will explore what Hydrocarbons Study Guide is, why Hydrocarbons Study Guide is vital, and how to effectively learn about Hydrocarbons Study Guide.
3. In chapter 2, this book will delve into the foundational concepts of Hydrocarbons Study Guide. This chapter will elucidate the essential principles that must be understood to grasp Hydrocarbons Study Guide in its entirety.
4. In chapter 3, this book will examine the practical applications of Hydrocarbons Study Guide in daily life. This chapter will showcase real-world examples of how Hydrocarbons Study Guide can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Hydrocarbons Study Guide in specific contexts. This chapter will explore how Hydrocarbons Study Guide is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Hydrocarbons Study Guide. The final chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Hydrocarbons Study Guide.

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